

ENGLISH FRIENDLY COURSES (EFC) 2023-2024 CAMPUS OF ÁLAVA

Link to website: https://www.ehu.eus/en/web/vitoria-gasteizko-ingeniaritza-eskola/incomingstudents

Contact: esc-ingenieria.internacional@ehu.es

In addition to the general offer of courses taught in English, some Centers offer for incoming students English Friendly Courses (EFC): subjects taught in Spanish or Basque, in which the syllabus summary; lecturer tutoring, examinations and/or papers are available in English.

Englis	sh Friendly Courses taught in SF				
	FACULTY OF ENGIN	NEERING - VITO	ORIA-GASTEI	Z (163)	
	COURSE	SEMESTER ¹	CREDITS	SCHEDULE ²	LINK TO SYLLABUS
Comm	on Courses for Industrial Branch				
25974	Fundamentos Físicos de la Ingeniería	Annual	12	М	
25975	Fundamentos Químicos de la Ingeniería	Annual	9	М	
25984	Mecánica Aplicada	Annual	9	А	
25980	Fundamentos de Tecnología Eléctrica	Annual	9	Α	
25979	Mecánica de Fluidos	2nd	6	А	
25985	Sistemas de Producción y Fabricación	2nd	6	Α	
Bache	lor's Degree in Industrial Electronics an	d Automation E	ngineering		
25999	Informática Industrial	1st	6	А	
26005	Sistemas Empotrados	1st	6	А	
28126	Cálculo y Diseño de Estructuras Automovilísticas	1st	6	Α	
28134	Aerodinámica	1st	6	М	
25996	Sistemas Electrónicos Digitales	2nd	6	М	
26006	Ampliación de Informática Industrial	2nd	6	А	
26007	Control por Computador	2nd	6	А	
28132	Simulación y Análisis FEM en Automoción	2nd	6	A	
Bache	lor`s Degree in Mechanical Engineering	5			
26045	Elasticidad y Resistencia de Materiales	1st	9	М	
26046	Cinemática y Dinámica de Máquinas	1st	9	Μ	

¹ SEMESTER: 1st: September 2023 to January 2024 2nd : January 2024 to May 2024

Annual: September 2023 to May 2024

² SCHEDULE: Morning (M)/ Afternoon (A): begins at 13.30.



	FACULTY OF ENGI	NEERING - VITC	ORIA-GASTEI	Z (163)	
	COURSE	SEMESTER ¹	CREDITS	SCHEDULE ²	LINK TO SYLLABUS
26050	Instalaciones y Máquinas Hidráulicas	2nd	6	М	
Bache	lor's Degree in Industrial Chemical Eng	ineering			
26091	Química Industrial	2nd	6	М	
26095	Gestión de residuos industriales	2nd	6	М	
Bache	lor's Degree in Automotive Engineering	g			
28138	Procesos de Fabricación en Tecnología Automotriz	1st	6	М	
28141	Automatización avanzada en fabricación de automóviles	1st	4,5	А	
28122	Introducción a la Mecánica	2nd	6	М	
Bache	lor's Degree in Computer Managemen	t and Informatio	n Systems Eng	gineering	
26018	Arquitectura de Computadores	1st	6	А	
26021	Lenguajes, Computación y Sistemas Inteligentes	1st	6	А	
26023	Investigación Operativa	1st	6	А	
26025	Sistemas de Gestión de Seguridad de Sistemas de Información	1st	6	М	
26031	Programación Básica	1st	6	М	
26036	Desarrollo de Sistemas Gráficos	1st	4,5	А	
26006	Ampliación de Informática Industrial	2nd	6	А	
26013	Metodología de la Programación	2nd	6	М	
26017	Ingeniería del Software	2nd	6	А	
26027	Sistemas de Apoyo a la Decisión	2nd	6	М	
26029	Sistemas Web	2nd	6	М	
26030	Administración de Bases de Datos	2nd	6	М	



English Friendly Courses taught in BASQUE:

	FACULTY OF ENGINEERING - VITORIA-GASTEIZ (163)									
	COURSE	SEMESTER ³	CREDITS	SCHEDULE ⁴	LINK TO SYLLABUS					
Bachel	lor's Degree in Automotive Engineeri	ing								
28140	Ibilgailu Elektrikoak eta Hibridoak	1st	6	М						
28122	Mekanikaren Oinarriak	2nd	6	М						

 ³ SEMESTER: 1st: September 2023 to January 2024
 2nd: January 2024 to May 2024
 Annual: September 2023 to May 2024
 ⁴ SCHEDULE: Morning (M)/ Afternoon (A): begins at 13.30.

Faculty 163 - Faculty of Engineering - Vitoria-Gasteiz	Cycle		
Degree GDIMEL10 - Doble Grado en Ingeniería Mecánica + Ingeniería Electrónica	Ind Year	 First year	
OURSE			
25975 - Chemical Fundamentals of Engineering	Cre	dits, ECTS:	9
OURSE DESCRIPTION			
Chemical Principles of Engineering enables students to acquire basic skills on the struct materials. This would make it possible to establish essential relationships between mate properties, as well as an understanding of certain industrial processes.			
Chemical Principles of Engineering is a compulsory subject (9 ECTS credits) taught at the Gasteiz in the first year of the following degrees: Bachelor in Industrial Electronic Engine Mechanical Engineering, Bachelor in Industrial Chemical Engineering, and the double bachenical Engineering + Business Administration and Management and in Industrial E Automatics + Mechanical Engineering.	eering and Aut achelor̵	omatics, Bac 7;s degrees i	helor
This subject serves as the basis for several subsequent subjects taught at the Bachelors Engineering and Automatics, and in Mechanical Engineering, such as Materials Science Technologies (4th year). With respect to the Bachelor in Industrial Chemical Engineering associated with this subject is of utmost importance regarding the subsequent tackling of Controlling and Instrumenting Chemical Processes, Experimentation in Chemical Engineering Chemical Reaction Engineering, Unit Operations or Analytical Chemistry.	e (2nd year) or g, the acquisiti of specific cour	Environmen on of the skil ses, such as	ls
OMPETENCIES/LEARNING RESULTS FOR THE SUBJECT			
COMPETENCES			
SPECIFIC COMPETENCES			
SPECIFIC COMPETENCES FB4. Ability to understand and apply the basic knowledge of principles of general chemis chemistry, and their applications in engineering.	istry, organic a	nd inorganic	
FB4. Ability to understand and apply the basic knowledge of principles of general chemis		-	
FB4. Ability to understand and apply the basic knowledge of principles of general chemis chemistry, and their applications in engineering.FB7. Ability to apply strategies of scientific method: analyze a problematic situation qual	litatively and q	uantitatively,	eerin
FB4. Ability to understand and apply the basic knowledge of principles of general chemistry, and their applications in engineering.FB7. Ability to apply strategies of scientific method: analyze a problematic situation qual propose hypotheses and solutions using appropriate models.FB8. Ability to communicate effectively the knowledge, procedures, results, skills and issues.	litatively and q	uantitatively,	eerin
 FB4. Ability to understand and apply the basic knowledge of principles of general chemistry, and their applications in engineering. FB7. Ability to apply strategies of scientific method: analyze a problematic situation qualipropose hypotheses and solutions using appropriate models. FB8. Ability to communicate effectively the knowledge, procedures, results, skills and iss subjects, using appropriate vocabulary, terminology and means. 	litatively and q sues relating to	uantitatively, o basic engin	eerin
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 FB4. Ability to understand and apply the basic knowledge of principles of general chemistic chemistry, and their applications in engineering. FB7. Ability to apply strategies of scientific method: analyze a problematic situation qualipropose hypotheses and solutions using appropriate models. FB8. Ability to communicate effectively the knowledge, procedures, results, skills and iss subjects, using appropriate vocabulary, terminology and means. TRANSVERSAL COMPETENCES FB10. Adoption of a responsible and ordered attitude at work, prepared to a lifelong lear LEARNING OUTCOMES 1. Know and apply models of the structure of the matter to understand the properties and 	litatively and q sues relating to rning experien ad behavior of s	uantitatively, o basic engin ce. substances a that chemica	Ind
 FB4. Ability to understand and apply the basic knowledge of principles of general chemistic chemistry, and their applications in engineering. FB7. Ability to apply strategies of scientific method: analyze a problematic situation qual propose hypotheses and solutions using appropriate models. FB8. Ability to communicate effectively the knowledge, procedures, results, skills and iss subjects, using appropriate vocabulary, terminology and means. TRANSVERSAL COMPETENCES FB10. Adoption of a responsible and ordered attitude at work, prepared to a lifelong lear LEARNING OUTCOMES 1. Know and apply models of the structure of the matter to understand the properties and materials (FB4). 2. Know and understand the basic principles and theories about the physical and chemic 	litatively and q sues relating to rning experien ad behavior of s cal processes be effects prod	uantitatively, o basic engin ce. substances a that chemica uced (FB4).	Ind
 FB4. Ability to understand and apply the basic knowledge of principles of general chemistic chemistry, and their applications in engineering. FB7. Ability to apply strategies of scientific method: analyze a problematic situation qualipropose hypotheses and solutions using appropriate models. FB8. Ability to communicate effectively the knowledge, procedures, results, skills and ississible subjects, using appropriate vocabulary, terminology and means. TRANSVERSAL COMPETENCES FB10. Adoption of a responsible and ordered attitude at work, prepared to a lifelong lear LEARNING OUTCOMES 1. Know and apply models of the structure of the matter to understand the properties and materials (FB4). 2. Know and understand the basic principles and theories about the physical and chemic substances can undergo under certain conditions in order to determine, in each case, the substances can undergo under certain conditions in order to determine, in each case, the substances can undergo under certain conditions in order to determine. 	litatively and q sues relating to rning experien ad behavior of s cal processes he effects prod ate lab-reports	uantitatively, o basic engin ce. substances a that chemica uced (FB4). (FB7).	Ind
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To achieve the objectives defined in the subject, the following selection of contents has been made:

Universidad Euskal Herriko del Pais Vasco Unibertsitatea

UNIT 1. BASIC PRINCIPLES

- Nomenclature of inorganic chemistry.
- States of matter.
- Basic concepts (atomic mass, mole concept, molar mass, etc.).
- Determination of chemical formulas (empirical and molecular formulas).
- Symbolic representation of chemical reactions through chemical equations.
- Basic concepts related with chemical equations (stoichiometry, limiting reagent, yield and purity).

UNIT 2. ATOMIC AND MOLECULAR STRUCTURE

- Quantum-mechanical model of the atom.
- Electronic structure.
- Periodic table. Periodic properties.
- Chemical bonding.
- Structure and properties of molecular compounds.

UNIT 3. AGREGATE STATES OF MATTER. PHASE EQUILIBRIA

- Solid state (types of crystalline solids and their characteristic properties (thermal conductivity, electrical conductivity in solid state, liquid or in solution, melting point and solubility, among others).

- Gaseous state (properties of the gases, ideal gaseous systems, kinetic theory of gases).
- Liquid state (surface tension, viscosity, vapour pressure).
- Equilibrium phase diagram.

UNIT 4. THERMOCHEMISTRY

- First principle of thermodynamics.

- Calculation of the change of internal energy (∆U) and enthalpy (∆H) in a substance. Energy exchange processes with and without phase change.

- Calculation of the change of internal energy (∆U) and enthalpy (∆H) in a chemical reaction.

UNIT 5. SOLUTIONS. COLLIGATIVE PROPERTIES

- Solutions and calculation of the concentration.
- Solubility and Henry's Law.
- Colligative properties.
- Volatile solute solutions. Ideal behaviour (Raoult's Law) and real behaviour (Pxy/Txy diagrams).
- -Simple distillation process.

UNIT 6. BASIC PRINCIPLES OF ORGANIC CHEMISTRY

- Nomenclature of organic compounds. Functional groups.

UNIT 7. KINETICS OF CHEMICAL REACTIONS

- Basic concepts of chemical kinetics: reaction rate, reaction order and reaction law.
- Simple kinetics models: zero, first and second order reactions.
- Effect of temperature. Arrhenius equation.
- Introduction to catalysis.

UNIT 8. CHEMICAL EQUILIBRIUM

- Thermodynamic principles of chemical equilibrium.
- Le Chatelier's principle.

UNIT 9. EQUILIBRIUM IN AQUEOUS SOLUTIONS

- Basic concepts (acid, base, conjugated species, amphoteric species).
- Acid (Ka) and basic (Kb) dissociation constant.
- pH scale.
- Reaction of hydrolysis.
- Acid-base titration (equivalence point, titration error and acid-base indicators).

UNIT 10. HETEROGENEOUS IONIC EQUILIBRIUM

- Precipitation reactions.
- Constant solubility product (Kps).
- The common-ion effect.
- The pH variation effect.
- Fractional precipitation.

UNIT 11. ELECTROCHEMISTRY

- Redox reactions.

- Voltaic or galvanic cell.
- Electrolytic cell.

TEACHING METHODS

PRESENTIAL ACTIVITIES

THEORETICAL CLASSES (45 h)

During the 30 weeks of the academic year, concepts and theoretical developments will be taught in a weekly session (1.5 h). Explanations will be complemented with standard exercises and activities that will allow the acquisition of stablished skills.

CLASSROOM PRACTICES (30 h)

During the 30 weeks of the academic year, resolution of exercises and practical activities will be carried out in a weekly session (1.0 h).

LABORATORY PRACTICES (15 h)

Students must complete 5 laboratory practices of 3 hours each. Lab sessions will be taught in the Laboratory practices will allow students to experiment and put into practice the knowledge acquired through lectures, classroom practices and personal work. Moreover, they will make it possible to learn about the basic experimental techniques used in a chemical laboratory and to acquire skills typical of laboratory work.

Students must complete 5 different laboratory practices of 3 h each. Lab sessions will be taught in the Basic Chemistry laboratory of the Faculty of Engineering Vitoria-Gasteiz, according to the calendar and schedule proposed for each group.

The practices will be carried out individually, as long as the available material allows it. Otherwise, the practices will be carried out in pairs, which will allow, additionally, to promote other skills, such as teamwork.

Each student must deliver a questionnaire before the beginning of each practical session, in which several questions related to the practice must be answered. This report must be answered and delivered individually at the time of entering the laboratory. At the end of the practical session, the students must take a test related to the content of the practice. Finally, the students will have one week to deliver a final report containing the results obtained and the main conclusions of the practice.

TUTORSHIP SESSIONS

In general, it is a voluntary activity (individual or collective) conducted in response to students' request. However, throughout the course a series of voluntary group deliverables will be proposed that will require attendance at tutorials.

NON-PRESENTIAL ACTIVITIES (135 h)

Continued work of student is essential to develop the competences of the subject. In addition to preparing the written exams, students should devote the hours of non-presential teaching to:

o Complete notes, consult bibliography and solve questions and/or problems, including voluntary deliverable tasks (a time commitment of approximately 3-4 h per week).

o Prepare the laboratory sessions (a time commitment of 1.5-2.0 h to prepare the laboratory practice and answer a set of preliminary questions per practice) and complete the corresponding report (2.0-3.0 h commitment per practice).

If the extraordinary circumstances derived from the current public health emergency situation caused by the outbreak of COVID-19 oblige to develop online teaching, all media available in the UPV/EHU (BlackBoard Collaborate, eGela, etc.) will be used.

	_		NA	_	•				TA		004	1
	I ypes Hours of face-to-fac	of teaching	M	S	GA	GL 15	GO	GCL	TA	TI	GCA	-
Horas de Activ	vidad No Presencial d		45 67,5		30 45	15 22,5						-
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Legend:	M: Lecture-based		-	Seminar					•••		n-based (
	GL: Applied laborato TA: Workshop	ry-based group			al worksł	ter-based	l groups		•••		based gro k groups	·
Evaluation m	•								, ipplied		it groupe	·
	us evaluation											
	ourse evaluation											
Evaluation to	ols and percentag	es of final I	nark									
- Written te	est, open questions	70%										
- Exercises	s, cases or problem											
	ry practices 20%											
-		lod: Guid	ELINE	ES ANI	D OPTI	NG OU	T					
EVALUATIO	ON SYSTEM											
WRITTEN	TEST/EXAM (70 %	of the final	mark)									
The written	test will comprise	wo partial te	ests:									
	est corresponds to binciding with the p										0	
	d test will be carrie y Call. In this case		, coin	ciding \	with the	period	establi	shed by	/ the Co	enter t	o perfor	m the exams of
If the stude month peric	nt has obtained a r od.	nark ≨	5;5 in 1	he first	test, h	e/she w	ill be ev	valuate	d on th	e conte	ents of t	he second fou
•	ade corresponding hark ≥4 has						the sim	ple ave	erage o	f the m	arks of	the two tests,
	nt has obtained a r he exam will be for											
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PRACTICA	L ACTIVITIES (10	% of the fina	al mar	k)								
	ctivities will be unde ires, amongst othe		ughou	t the co	urse, s	uch as	problen	n solvir	ig and o	cases,	written	tests or
LABORATO	ORY PRACTICES	(20 % of the	final r	nark)								
 Presentation Evaluation 	y work: 25 % ion and evaluation n test (after each la ion and evaluation	b-session): 2	25 %			%						
REQUIREN	/IENTS to pass the	subject:										
	all the leberatory		l daliv	er all th	e nrevi	0118 0116	estionn	aires, p	ost-pra	ctice e	valuatio	on tests and th

Universidad Euskal Herriko del País Vasco Unibertistatea o Obtain a mark ≥5 out of 10 in the written test (70 %).

o Obtain a mark ≥4 out of 10 in the laboratory practices (20 %).

o It is not necessary to obtain a minimum mark in the practical activities (10 %).

Those students who do not meet any of these requirements will be marked with a 4.0 (maximum) in the Ordinary Call regardless of the final grade obtained.

*If the extraordinary circumstances derived from the current public health emergency situation caused by the outbreak of COVID-19 oblige to develop an online evaluation, all media available in the UPV/EHU (Webex, eGela, etc.) will be used. The characteristics of this new evaluation mode will be published in eGela.

CALL RESIGNATION

Those students who do not appear for the written test will be recorded as "Not Presented" in the Ordinary call.

FINAL TEST

Students who meet the conditions established in the UPV/EHU regulations and request to take a final test within the deadline set for that purpose (Chapter II, Article 8 of the Agreement of December 15, 2016, of the Governing Council of the University of the Basque Country/Euskal Herriko Unibertsitatea, which approves the Regulations governing the students' Evaluation in official Bachelor's degrees), they need to implement the following activities:

- A written test related to the theoretical-practical contents of the subject (80 % of the final grade).

- A practical laboratory exam (20 % of the final exam).

REQUIREMENTS to pass the subject (FINAL TEST)

Obtain a mark equal to or greater than 5 in the final grade (obtained as the weighted average of the marks corresponding to the written test and the practical exam).

*If the extraordinary circumstances derived from the current public health emergency situation caused by the outbreak of COVID-19 oblige to develop an online evaluation, all media available in the UPV/EHU (BlackBoard Collaborate, eGela, etc.) will be used. The characteristics of this new evaluation mode will be published in eGela.

CALL RESIGNATION

Those students who do not appear for the final test will be recorded as "Not Presented" in the Ordinary call.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

The evaluation criteria in the extraordinary call will be:

- A written test formed by the following characteristics:

If the student has obtained a mark ≥5 in the exam related to the theoretical-practical contents of the first fourmonth period (mark obtained in the Call of January or May, with a weight of 40 % of the final grade of the Extraordinary call) and <5 in the exam related to the theoretical-practical contents of the second four-month period (mark obtained in the call of May), in the exam of the Extraordinary Call the student will be evaluated on the contents of the second fourmonth period. In this case, the exam will have a weight of 40 % of the final grade.

The final grade corresponding to the written test will be calculated as the simple average of the marks of the two tests, as long as a mark ≥4 has been obtained in the Extraordinary call exam.

If the student has obtained a mark <5 in the exam related to the theoretical-practical contents of the first four-month period (mark obtained in the Call of January or May) and ≥5 in the exam related to the theoretical-practical contents of the second four-month period (mark obtained in the call of May, with a weight of 40 % of the final grade of the Extraordinary call), in the exam of the Extraordinary Call the student will be evaluated on the contents of the first four-month period. In this case, the exam will have a weight of 40 % of the final grade.

The final grade corresponding to the written test will be calculated as the simple average of the marks of the two tests, as long as a mark ≥4 has been obtained in the Extraordinary call exam.

If the student has obtained a mark <5 in the exams related to the theoretical-practical contents of both four-month periods (both in the January and May calls), in the exam of the Extraordinary Call the student will be evaluated on the theoretical-practical contents of the whole subject. In this case, the exam will have a weight of 80 % on the final grade.

The final grade corresponding to the written test will be calculated as the simple average of the grades obtained in the 2 parts that make up the exam of the Extraordinary Call, being necessary that both grades are ≥4 out of 10.

- A practical laboratory exam (20 % of the final exam). A student will be exempt from this exam if all the laboratory practices throughout the course are completed and a mark ≥4 is obtained; this grade will have a weight of the 20 % of the final grade.

REQUIREMENTS to pass the subject

Obtain a mark ≥5 in the final grade (obtained as the weighted average of the marks corresponding to the written test and the practical exam or laboratory practices).

*If the extraordinary circumstances derived from the current public health emergency situation caused by the outbreak of COVID-19 oblige to develop an online evaluation, all media available in the UPV/EHU (BlackBoard Collaborate, eGela, etc.) will be used. The characteristics of this new evaluation mode will be published in eGela.

CALL RESIGNATION

Those students who do not appear for the written test will be recorded as "Not Presented" in the Extraordinary call.

MANDATORY MATERIALS

Collections of problems and specific questions related to the subject.

BIBLIOGRAFÍA

Basic bibliography

NOMENCLATURE OF INORGANIC CHEMISTRY

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o Quiñoá, E.; Riguera, R.; Vila, J. M. (2010). Nomenclatura y formulación de los compuestos inorgánicos. Ed. McGraw-Hill.

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GENERAL CHEMISTRY

o Brown, T. D.; Lemay, H. E.; Bruce, J. R.; Bursten, E.; Burdge, J. (2003). Química. La Ciencia Central. Ed. Pearson Prentice Hall.

o Casabó, J. (2009). Egitura atomikoa eta lotura kimikoa. Ed. UPV/EHU.

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o McMurry, J.E.; Fay, R.C. (2009). Química General. Ed. Pearson Education, 5º edición.

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LABORATORY PRACTICES

o Beran, J.A. (2014). Laboratory Manual for Principles of General Chemistry. Ed. Wiley.

o Chemical Education Material Study (1987). Química. Una ciencia experimental. Ed. Reverté.

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o Navarro, A.; Gonzalez, F. (1986). Prácticas y técnicas de laboratorio. Ed. Universidad Politécnica de Catalunya.

Detailed bibliography

o Casabó, J. (2007). Estructura atómica y enlace químico. Editorial Reverté.

o Ghasem, N.; Henda, R. (2009). Principles of Chemical Engineering Processes. Editorial Taylor & Francis Group.

o Primo-Yúfera, E. (2007). Química Orgánica básica y aplicada: de la molécula a la industria. Tomo II. Editorial Reverté.

o Rodgers, G. E.; Cabañas, M. V.; Regi, M. V. (1995). Química inorgánica: introducción a la química de coordinación, del estado sólido y descriptiva. Editorial McGraw-Hill Interamericana de España.

Journals

Web sites of interest

http://www.egela.ehu.es

http://www.ptable.com/?lang=es

https://www.luc.edu/media/lucedu/sustainability-new/pdfs-biodiesel/Biodiesel%20Curricula%20-%20Version%205.0.pdf

OBSERVATIONS

Faculty 163 - Faculty of Engineering - Vitoria-Gasteiz	Cycle		
Degree GIAUTO10 - Bachelor's Degree in Automotive Engineering - DUAL	Year	First year	
OURSE			1
25979 - Fluid Mechanics	С	redits, ECTS:	6
OURSE DESCRIPTION			
Fluid Mechanics is a key subject of the Second grade in Industrial Engineering in the L Vitoria-Gasteiz with 6 ECTS credits.	Iniversity Coll	ege of Enginee	ering a
Fluid Mechanics deals with the study of all fluids under static and dynamic situations. F continuous mechanics which deals with a relationship between forces, motions, and sta material. This study area deals with many and diversified problems such as surface ter bodies, or flow round bodies (solid or otherwise), flow stability, etc.	atical conditio	ns in a continu	ous
Fluid mechanics is widely used both in everyday activities and in the design of modern cleaners to supersonic aircraft. Therefore, it is important to develop a good understand Mechanics.		-	
An ordinary house is, in some respects, an exhibition hall filled with applications of Flui for cold water, natural gas, and sewage for an individual house and the entire city are of Fluid Mechanics. The same is also true for the piping and ducting network of heating a refrigerator involves tubes through which the refrigerant flows, a compressor that press exchangers where the refrigerant absorbs and rejects heat. Fluid mechanics plays a m components.	designed prim nd air-conditio surizes the ref	arily on the bas oning systems. rigerant, and ty	sis of A wo hea
All components associated with the transportation of the fuel from the fuel tank to the of fuel injectors, or carburetors as well as the mixing of the fuel and the air in the cylinders gases in exhaust pipes are analyzed using fluid mechanics. Fluid mechanics is also us air-conditioning system, the hydraulic brakes, the power steering, automatic transmissi cooling system of the engine block including the radiator and the water pump, and even	s and the purg ed in the des on, and lubric	ging of combus gn of the heati	tion ng an
On a broader scale, fluid mechanics plays a major part in the design and analysis of ai jet engines, wind turbines, biomedical devices, the cooling of electronic components, a crude oil, and natural gas. It is also considered in the design of buildings, bridges, and the structures can withstand wind loading. Numerous natural phenomena such as the of ground water to the top of trees, winds, ocean waves, and currents in large water bor principles of Fluid Mechanics.	nd the transp even billboard rain cycle, we	ortation of wate ds to make sure ather patterns,	er, e that the ri
The students of the Grade in Industrial Chemical Engineering will apply the knowledge year, such as Physical Chemistry, Control of Chemical Processes and Experimentation students of the Grade in Engineering in Automotive, will later apply the knowledge acquared Aerodynamics, third year.	n in Chemical	Engineering I.	
OMPETENCIES/LEARNING RESULTS FOR THE SUBJECT			
The following course skills are developed:			
 Knowledge of basic and technological subjects that enables students to learn new me with versatility to adapt to new situations), Capacity to solve problems using initiative, decision making, creativity, critical thinking knowledge, abilities and skills in the field of Industrial Engineering, and the cross-curric 	g, and to com	municate and o	-
- Adopt a responsible and organised attitude towards work and a willingness to learn ta the necessary continuous training,			nge of
- Apply scientific method strategies: analyse qualitatively and quantitatively the problem and solutions using industrial engineering models, speciality mechanics, and	n situation, pr	opose hypothe	ses
 Work efficiently in a group, integrating skills and knowledge to make decisions in the the the competencies and key knowledge that this course programme offers can be used grade in Industrial Engineering: Hydraulic machinery Hydraulic installations Pneumatic and hydraulic systems 			-

Universidad Euskal Herriko del Pais Vasco Unibertsitatea

CONTENIDOS TEÓRICO-PRÁCTICOS

In order to get the background knowledge, abilities and skills, the course content is divided into five blocks of learning units: Hydrostatics, Kinematics and Dynamics, Dimensional analysis, similitude and viscous flows, Flow Hydraulic machinery and Installations in pipes and open channels

Theoretical content (chapters):

- 1. Fluid Mechanics. Basic concepts.
- 2. Fuid Properties. Basic definitions.
- 3. General laws of the Fluid Statics.
- 4. Pneumatics and hydraulics. Basic concepts.
- 5. Pneumatic and hydraulic circuits.
- 6. Statics of incompressible fluids in the gravitational field. Hydrostatics.
- 7. Fluid forces on surfaces.
- 8. Fluid forces on submerged and floating bodies.
- 9. Fundamentals of Fluid Kinematics.
- 10. Mass conservation theorem. The continuity equation.
- 11. Fundamental equation of Fluid Dynamics.
- 12. Bernoulli equation.
- 13. Applications of the Bernoulli equation. Flow meters.
- 14. Momentum equation. Angular momentum equation.
- 15. Applications of the momentum equation.
- 16. Dimensional analysis and dynamical similitude.
- 17. Viscous flows.
- 18. Head losses in pipes.
- 19. Steady flow in conduits and pipes. Multipath pipelines problems.
- 20. Varying flow in pipes.
- 21. Steady flow in open channels.
- 22. Hydraulic machinery. Fundamentals. Turbo machinery.
- 23. Hydraulic turbines. Hydro electrical stations/plants.
- 24. Hydraulic pumps.
- 25. Pumping installations/stations.

Practical content:

The students will perform 17 or 18 laboratory experiments.

1. Measurement of viscosity of a fluid

- 2. Rigid-body rotation of fluids
- 3. Fluid forces on surfaces
- 4. Verification of Bernoulli equation
- 5. Discharge in tanks
- 6. Study of Flow meters
- 7. Use of Weirs
- 8. Forces exerted by fluid jets
- 9. Study of primary (friction) head losses in pipes.
- 10. Study of secondary (minor) head losses in pipes.
- 11. Determination of cavitation in pipes.
- 12. Analysis of Pelton turbines
- 13. Analysis of centrifugal pumps
- 14. Centrifugal fan.
- 15. Time to empty a tank containing a liquid.
- 16. Water hammer
- 17. Wind tunnel
- 18. Pneumatics and hydraulic systems

Required materials (background, instructions, and lab report template) are posted on the course. Students will be assigned into groups/teams formed by three students, to perform the experiments. Group lab reports will be submitted after finish the lab.

Attendance to the laboratory sessions is compulsory.

The course is geared towards self-learning and uses participatory approaches as much as possible. A cooperative learning (AC, Aprendizaje cooperativo) methodology will be used, including lecture/presentation, group work, demonstrations, case studies, problem solving practical sessions (hands-on practice), small and large group exercises and role plays. The attendees roles and responsibilities will be change in the group/team.

The course objectives are:

- To introduce definitions, concepts, properties, principles, laws, observations and models of ideal and real fluids at rest and in motion.

- To provide basis for understanding fluid behavior at rest and in motion (laminar, turbulent) and for engineering design and control of fluid systems.

- To develop competence with mass, energy and momentum balances for determining resultant interactions of flows and engineered and natural systems.

- To develop basis for correlating experimental data, designing tests, and using scale models of fluid flows.

- To learn nature of rotation, circulation, resistance (viscous, turbulent), boundary layers and separation with applications to drag and lift on objects

- To learn methods for computing head losses (friction and fitting losses) and flows in simple pipes and channels.

- To learn the fundamentals of pumps and hydraulic turbines and the way they operate.

- To identify and understand how the key elements work: compressor, cooler, separator, actuators, valves and accessories involved in pneumatic and hydraulic facilities.

- To solve engineering problems associated with pneumatic and hydraulic installations, designing a series of practical circuits.

- To contribute primarily to the students' knowledge of college-level mathematics and/or basic sciences and provide experimental experience.

The student will be able:

- To interpret, define and solve practical problems related with the nature of different types of fluids and their interactions on engineered and natural systems in order to develop technical projects.

- To identify, interpret and explain the terminology, the structural characteristics, key parts, operation and application fields of pumps and hydraulic turbines and manage that knowledge to choose the suitable machine for every installation, according to technical criteria.

- To prepare, present, defend, orally and in writing, and make reports on the subject working individually or in groups.

- To analyze, interpret and synthesize a Technical Project related to Fluid Mechanics.

The laboratory experiments, team project and homework assignments will be performed in groups in cooperative work.

A team project titled "Design and calculation of a pumping installation" will be performed in groups. The student group will have to identify and set all the parameters involved in the project according to the instructions provided by the teachers.

The student groups will co evaluate the work made by the rest of the groups as for instance the team project.

The following individual or group assignments will be made along the course:

- Initial opinion survey on the subject
- Group/team meeting minutes
- 5/6 homework assignments
- Individual pop quizzes
- A mid-term exam
- Project Design sheet (planning sheet)
- First part of the Project Report
- Final Project Report

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- Public presentation of the project

An evaluation will be expected at the end of the semester for students to give feedback on the course, and to outline what they have learned based on:

- Evaluation sheet

- Final Opinion survey

TYPES OF TEACHING S GL GO GCL TA ΤI GCA Types of teaching Μ GA Hours of face-to-face teaching 45 15 Horas de Actividad No Presencial del Alumno/a 67,5 22,5 Legend: M: Lecture-based S: Seminar

GL: Applied laboratory-based groups TA: Workshop

GO: Applied computer-based groups

TI: Industrial workshop

GA: Applied classroom-based groups GCL: Applied clinical-based groups GCA: Applied fieldwork groups

Evaluation methods

- End-of-course evaluation

Evaluation tools and percentages of final mark

- Written test, open questions 30%
- Exercises, cases or problem sets 30%
- Teamwork assignments (problem solving, Project design) 30%
- Oral presentation of assigned tasks, Reading; 10%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

The course's assessment will be continuous, based on two mid-term examinations, weekly homework assignments, pop quizzes, and the realization of a laboratory report and a technical project based on the design of a pumping system.

Homework assignments: 20 % Team work Mid term exam: 30 % Chapters: 1-17. Laboratory Report: 15 % Team work Technical Project: 35 % Team work

More specifically explained:

20%. Submission and assessment of homework assignments (deliverables or tasks assigned for the different topics). When students submit less than 80% of homework assignments, they will receive the grade Not submitted. The quality of the group assignment will also be taken into account.

30%. Students will take a mid term exam prior to carrying out the project.

15%. Report on laboratory practice, carried out in groups/teams. Completion of lab practice and corresponding report is compulsory in order to pass the subject. The quality of team work will also be taken into account.

35%. Completion of a Team Technical Project. The project evaluation will be function of the oral project presentation (presentation depending on the number of students enrolled in the academic course), the quality of the team work performed, as well as the individual evaluation tests and/or groups that are held during the last 5 weeks of the course.

- To pass the course, students are required to pass the two mid term exams.

- Students failing the mid-term exams will have the choice to pass a retake exam in May.

- Only the students fulfilling the requirements of the Official College Regulations (Article 43.1, Section c) will have the right to take a final ordinary exam.

- The students not attending to class or laboratory sessions or project classes or submitting less than 80% of the homework assignments will be Not Evaluated.

- There will be a final ordinary exam at the end of the course semester (in May) for those students who failed to pass the course by the previously explained evaluation method.

SOME REMARKS:

⁻ Final exam in June's call: for the student who does not pass the subject by continuous assessment. 100% of the mark.

For students to justify the impossibility of continuous evaluation in the direction of the School:

- Final exam (call for May and June). 100% of the mark.

Article 43 of the Management Regulations for the teaching of undergraduate and first and second cycle, provides the proper reasons for non-participation in the continuous assessment (work reasons, victims of domestic violence, birth, adoption, foster care or daughters and children under three years in charge, care of dependent family member, students with disabilities equal to or greater than 33%, high-level athlete, artistic / cultural activities that involve travel or dedication, compatibility with other higher education, compatibility with political office , union, student representation, associations, NGOs, or other).

Once the student has taken part in a partial continuous assessment tests, it is assumed that you are following the continuous evaluation and get a final score calculated by weighting all tests. ONLY if they are not present at any of the tests, you get a rating of "not presented".

Students not performing the Laboratory experiments or the Project will receive the mark of NOT PRESENTED in the corresponding call.

Students who fail to pass the course according to the previously explained system of continuous evaluation will have the choice to take a final exam in the corresponding ordinary and extraordinary calls (June, July).

The students who decide not to follow the course according to the previously explained system of continuous evaluation will notify the waiver (renunciation) of continuous assessment to the teacher, and they will have the choice to take a final exam in the regular examination calls where all competencies and learning outcomes identified will be evaluated.

The waiver or renunciation of continuous evaluation may be applied during the teaching period of the subject. In any case, students which are not able to attend class on a regular basis because they are working or complying with the requirements of the management regulations for the first and second cycle courses, are asked to contact the teaching staff for an adapted program of development of competences and learning objectives of the subject.

Students who do not participate in the exams and/or in the project and/or in the laboratory practices, will receive the qualification of Not Presented in the corresponding call.

The final exam will be the same for all the groups.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

- The student failing the course will also have the choice to take a final extraordinary exam in June based on all the contents and problems studied all along the course.

MANDATORY MATERIALS

Most part of the following teaching material will be available on E-gela:

Teacher resource notes Tables and diagrams Laboratory manual PowerPoint presentations (slides) Solved exams Problem statements and solutions Student guide, course project guide Appendices (minutes, forms, sheets) Homework assignments

BIBLIOGRAFÍA

Basic bibliography

Most part of the bibliography listed for the subject and more can be found on the signature 532 in the Biblioteca de las Nieves library.

Fluid Mechanics. Fundamentals and Applications. Cengel, Y.A, J.M. Cimbala. 2nd Edition in SI units. Mc Graw Hill. 2009. Signature: 532CEN (Sala Font Quer).

Fluid Mechanics. F.M. White. 7th Edition. Mc Graw Hill, 2011. ISBN: 978-007-131121-2.

Mecánica de Fluidos Incompresibles y Turbomáquinas Hidráulicas. Agüera Soriano, José. 5ª Edición. Edit. Ciencia. 2002.

Mecánica de los Fluidos, V.L. Streeter, E.B. Wylie. 9th Edition Mc Graw Hill, 2000. ISBN: 958-600-987-4.

Introducción a la Mecánica de Fluidos. Fox, R.W. McDonald, A.T. 4ª Edition McGraw-Hill. 1995.

Fundamentos de Mecánica de Fluidos. Gerhart, P.M, Gross, R.J. y Hochstein, J.I., 2ª Edición. Ed. Addison-Wesley Iberoamericana S.A. 1995.

Fox, R.W. McDonald, A.T. Introducción a la Mecánica de Fluidos. 4ª Ed. McGraw-Hill. 1995.

Gerhart, P.M, Gross, R.J. y Hochstein, J.I. Fundamentos de Mecánica de Fluidos, 2ª edición. Ed. Addison-Wesley Iberoamericana S.A. 1995.

Detailed bibliography

Introductory Fluid Mechanics. J. Katz. Cambridge University Press. 2010. ISBN: 978-0-521-19245-3.

Mechanics of Fluids. Potter, M., D. Wiggert. 3rd Edition. Thomson, 2002. ISBN: 970-686-205-6.

Applied Fluid Mechanics. R.L. Mott. 6th Edition. Pearson Prentice Hall, 2006. ISBN: 0-13-197643-5.

Mecánica de Fluidos. Shames, Irving H. Mc Graw-Hill. 1995.

Fluid Mechanics with Engineering Applications. Finnemore, E.J. y Franzini, J.B. 2002.

Engineering Fluid Mechanics. Crowe, Elger, Williams and Roberson. 9th Edition. Wiley & Sons, Inc. 2010. ISBN: 978-0-470-40943-5.

Journals

- Computers and Fluids
- El instalador
- Environmental Fluid Mechanics
- European Journal of Mechanics. Series B. Fluids
- Experimental Thermal and Fluid Science
- Experiments in Fluids
- Flow Measurement and Instrumentation
- Fluid Dynamics Research
- Fluidos
- Geophysical and Astrophysical Fluid Dynamics
- Ingeniería Del Agua
- International Journal of Multiphase Flow
- International Journal of Heat and Fluid Flow
- International Journal of Heat and Mass Transfer
- Journal of Fluid Mechanics.
- Journal of Fluids Engineering
- Journal of Hydraulic Engineering
- Journal of Non-Newtonian Fluid Mechanics
- Montajes e instalaciones
- Physicochemical Hydrodynamics
- Physical review A. Statistical physics, plasmas, fluids, and related interdisciplinary topics
- Physical review E. Statistical physics, plasmas, fluids, and related interdisciplinary topics
- Physics of fluids
- Physics of fluids A. Fluid Dynamics

- Tecnología del agua

Web sites of interest

- Hydraulic Institute. www.pumps.org
- Pump-Flo Co. www.pump-flo.com/manulist.asp
- Animated software company, www.animatedsoftware.com
- Pumps and systems magazine: www.pump-zone.com.
- http://www.sc.ehu.es/sbweb/fisica/fluidos/estatica/introduccion/Introduccion.htm.
- National Committee for Fluid Mechanics Films (NCFMF) http://web.mit.edu/fluids/www/ Shapiro/ncfmf.html
- IIHR- Hydroscience & Engineering, College of Engineering, The University of Iowa. http://www.iihr.uiowa.edu/:
- Enciclopedia básica sobre fluidos: http://hyperphysics.phy-astr.gsu.edu/ hbase/ fluid.html#flucon
- Principios de aeronáutica: http://wings.avkids.com/Libro/advanced.html
- Simulación de redes de distribución de fluidos: http://www.epa.gov/nrmrl/wswrd/dw/epanet.html
- UNESCO-IHE Institute for Water Education: http://www.unesco-ihe.org/
- Manual de vuelo: http://www.inicia.es/de/vuelo/

OBSERVATIONS

Tutorials: The students have a schedule of tutorials to deal with all issues related to the subject. Its use is encouraged to support the acquisition of the competences of the subject with the close support of the teacher, who is available to attend and help students. Outside the tutoring hours there will be no problem in attending students, whenever possible. It is recommended by appointment.

COURSE GUIDE	2023/24			
Faculty 163 - Faculty of	Engineering - Vitoria-Gasteiz	C	ycle .	
Degree GIEIAU10 - Bad	chelor's Degree in Industrial Electronics and	Automation Engine Ye	ear Second y	ear
COURSE				
25984 - Applied Mechanics			Credits, ECTS:	9
COURSE DESCRIPTION				1
DESCRIPTION AND CONT	EXTUALIZATION OF THE SUBJECT.			
rigid solids. These ideas will serve as a Applied Mechanics has a ve sense of the students. The o once the parts are understoo of vector calculus and matrix	scientific-technical basis for the engineers in ery close relationship with Physics and Mathe course will serve to develop the analytical ab od to be able to solve the problem as a whole x algebra, so the skills acquired in the Physic re the problems numerically and symbolically	the industrial area. ematics, and will serve wility to divide a problem e. The concepts of this cs, Calculus and Algeb	to broaden the phys n into simpler parts, s subject are within t	sical so tha
COMPETENCIES/LEARNING	RESULTS FOR THE SUBJECT			
COMPETENCES / LEARNII	NG OUTCOMES OF THE SUBJECT.			
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Kinematics of systems comp Topic 5 DYNAMICS Laws of Topic 6 INTRODUCTION TO	cept of position, velocity and acceleration. Re posed of rigid solids. f dynamics. Theorems of dynamics, for the p O THE STUDY OF THE RESISTANCE OF M Cortadura and its applications. General theor	point and for material sy MATERIALS Tension a	ystems.	
TEACHING METHODS		,		
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Universidad Euskal Herriko del Pais Vasco Unibertsitatea

	Types of teaching	М	S	GA	GL	GO	GCL	ТА	TI	GCA	
	Hours of face-to-face teaching	60		30							
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Legend:	M: Lecture-based	S:	Seminar				GA: A	pplied c	lassroon	n-based gro	ups
	GL: Applied laboratory-based grou	•		d compu		d groups		•••		based group	S
	TA: Workshop	TI:	Industria	al worksh	юр		GCA:	Applied	fieldwor	k groups	
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- BILBAO, A; AMEZUA, E. Mecánica Aplicada: Estática y cinemática. Edit: Síntesis.
- BILBAO, A; AMEZUA, E; ALTUZARRA, O. Mecánica Aplicada: Dinámica. Edit: Síntesis.
- BASTERO, J.M.; CASELLAS, J. Curso de Mecánica Edit: EUNSA
- Manuel Vazquez. Resistencia de Materiales. Editorial: Universidad Politécnica de Madrid
- Luis Ortiz Berrocal. Resistencia de Materiales. Editorial Mc Graw Hill
- Timoshenko. Resistencia de Materiales (2 tomos). Editorial: Espasa-Calpe

Journals

Web sites of interest

http://www.vc.ehu.es/ingme

http://egela.ehu.eus

http://www.biblioteka.ehu.eus

OBSERVATIONS

COURSE GUIDE	2023/24			
Faculty 163 - Facu	ulty of Engineering - Vitoria-Gasteiz	Cycle		
Degree GIEIAU10) - Bachelor's Degree in Industrial Electronics and Automation	n Engine Year	Second y	ear
OURSE				
25985 - Production and	d Manufacturing Systems	Cre	dits, ECTS:	6
OURSE DESCRIPTION				
metal–mecánic	los fundamentos del conocimiento y la aplicación de los proc a en una secuencia lógica de acuerdo a su ubicación en el c signatura, y el gran número de interrelaciones con otros car	contexto productivo.	De ahí el ca	ráctei
COMPETENCIES/LEAR	NING RESULTS FOR THE SUBJECT			
Técnicas, equipos y pr	rocesos de producción. Sistemas de fabricación flexible			
heoretical and Practica	al Contents			
Diseño de procesos de Distribución en planta Gestión de stocks de o Planificación de ventas	demanda independiente s y operaciones a de la producción (MPS - Master Production Scheduling) ements Planning)	[·] undición, soldadura	y otros	
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Las clases MAGISTRALES servirán para exponer los fundamentos teóricos de los procesos de fabricación y de a materias que están íntimamente ligadas al desarrollo y control de los mismos, tratando de describir las múltiples relaciones existentes con otras disciplinas y con la aplicación de los conocimientos que se adquirirán en otras

especialidades de la Ingeniería.

Asimismo, se propondrá la realización de ejercicios o problemas que se recogerán en clase. Los ejercicios se entregarán individualmente.

Además, se propondrán trabajos. Éstos se realizarán preferentemente en parejas y se expondrán en las últimas semanas de clase de la asignatura.

TYPES OF TEACHING

Hours of face-to-face teaching 45 15 Image: Constraint of the sector of the sect		Types of teaching	М	S	GA	GL	GO	GCL	TA	TI	GCA
Legend: M: Lecture-based S: Seminar GA: Applied classroom-based GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-based groups TA: Workshop TI: Industrial workshop GCA: Applied fieldwork groups	Hour	s of face-to-face teaching	45		15						
GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-based groups TA: Workshop TI: Industrial workshop GCA: Applied fieldwork grou	loras de Actividad	No Presencial del Alumno/a	67,5		22,5						
TA: Workshop TI: Industrial workshop GCA: Applied fieldwork grou Evaluation methods Evaluation methods	Legend: M: L	ecture-based	S: :	Seminar				GA: A	pplied c	lassroom	n-based g
Evaluation methods	GL:	Applied laboratory-based grou	ps GC): Applie	d compu	ter-base	d groups	GCL:	Applied	clinical-b	based gro
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	valuation methor	ls									
- End-of-course evaluation	- End-of-course	evaluation									
valuation tools and percentages of final mark	aluation tools a	nd percentages of final	mark								
- Written test, open questions 70% - Exercises, cases or problem sets 15% - Oral presentation of assigned tasks, Reading¿ 15%	- Exercises, cas	es or problem sets 15%		, 15%							
ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT	RDINARY EXAM	NATION PERIOD: GUIE	DELINE	ES AND	OPTII	NG OU	Т				
- EXÁMENES 80%											

- ENTREGABLES (CUESTIONES, PROBLEMAS, TRABAJOS...) 20%

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

En segunda convocatoria se realizará una única prueba o examen final (prueba escrita) en el que se contemplarán todos los contenidos desarrollados durante la asignatura.

Si el estudiante no se presenta al examen de la convocatoria EXTRAORDINARIA (junio/julio), obtendrá un "No presentado" independientemente de su participación en la realización de prácticas o en la exposición de trabajos.

MANDATORY MATERIALS

Apuntes y documentación proporcionada por el profesorado de la asignatura.

BIBLIOGRAPHY

Basic bibliography

Diseño de Procesos de Producción y distribución en planta:

Suñe, A., Gil, F. y Arcusa, I. (2005) Manual práctico de diseño de sistemas productivos. Díaz de Santos. Alarcón Valero y otros. (2004) Diseño de sistemas productivos y Logísticos. Universidad Politécnica de Valencia. Cuatrecasas, Ll., (1998). Gestión competitiva de stocks y procesos de producción. Gestión 2000. Cuatrecasas, LL (2000). Diseño de procesos de producción flexible. TGP.

Gestión de stocks:

Fernández, M. Planificación y gestión de la producción. ICAI Krajewski, L.; Ritzman, L. (2000) Administración de operaciones. Estrategia y análisis. Prentice Hall. Schroeder, Roger. (2005) Administración de operaciones. Concepto y casos contemporáneos. McGrawHill.

Planificación de Ventas y Operaciones:

Vollmann, T. y otros. (2005) Planeación y control de la producción. Administración de la cadena de suministros. McGrawHill.

Plan Maestro de Producción:

Krajewski, L.; Ritzman, L. (2000) Administración de operaciones. Estrategia y análisis. Prentice Hall.

MRP y MRP II:

Cuatrecasas, Ll., (1998). Gestión competitiva de stocks y procesos de producción. Gestión 2000. Vollmann, T. y otros. (2005) Planeación y control de la producción. Administración de la cadena de suministros. McGrawHill.

JIT:

Marin, F: Delgado, J. (2000). Las técnicas justo a tiempo y su repercusión en los sistemas de producción. Revista Economía Industrial. Nº 331

Programación y control producción:

Alfaro, J.; Pérez, D.; García, J.; Andrés, C.; Lario, F.C. (2002). Problemas de programación y control de producción. Universidad Politécnica de Valencia.

Andrés, C. y otros. Apuntes de programación y control de producción

Detailed bibliography

Journals

Web sites of interest

http://ekasi.ehu.es/ http://moodle.ehu.es/moodle/

OBSERVATIONS

Faculty Degree				Cyclo		
_	-	of Engineering - Vitor		Cycle		
CUIDEE	GIEIAU10 - Ba	chelor's Degree in I	ndustrial Electronics and Automation	Engine Year	Third yea	r
COURSE						
25996 - Dig	ital Electronic S	ystems		0	Credits, ECTS:	6
COURSE DES	CRIPTION					
systems the The referred utility. It is conven	eir most commo d devices are in ient to have dev y face this subje	n peripherals and pro the core of almost a veloped the compete	Isory subject aiming at introducing th otocols. ny electronic equipment manufacture ncies acquired in Digital Electronics a the starting point for the following sub	ed today which and Informatics	portrays its prac	ctical to
COMPETENC	IES/LEARNING	RESULTS FOR TH	E SUBJECT			
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Interrupts a 8051 asser Asynchron Synchrono I/O periphe	ous communica	C and SPI / arrays	S-485			
Interrupts a 8051 asser Asynchrono Synchrono I/O periphe Voltage su Other archit	ous communica us protocols: I2 erals: LCDs, Key pervisor and wa tectures	C and SPI / arrays itchdogs				
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The student should be able to handle different bibliographic resources and datasheets from manufacturers.

Universidad Euskal Herriko del Pais Vasco Unibertsitatea

TYPES OF TEACHING S GL GO GCL TA ΤI GCA Types of teaching Μ GA Hours of face-to-face teaching 45 15 Horas de Actividad No Presencial del Alumno/a 67,5 22,5 Legend: M: Lecture-based S: Seminar GA: Applied classroom-based groups GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-based groups TA: Workshop TI: Industrial workshop GCA: Applied fieldwork groups **Evaluation methods** - Continuous evaluation - End-of-course evaluation Evaluation tools and percentages of final mark - Multiple choice test 70% - Individual assignments 30% **ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT** Mounting a working microcontroller based PCB is compulsory to be evaluated. The assessment work will be presented to the professor who will evaluate the specifications fulfillment (25%), the right use of the explained methodologies (10%), the grade of self-development (15%) and the exposition (10%). The assessment can be carried out from the 11th week. Tests (40 %) will be carried out before starting with the microcontroller project. The mark obtained in this case will be the average of the tests marks. In order to renounce this call the student only needs not to hand in the expected microcontroller work. If the student prefers to take a final exam, it will account for 100 % of the note. To be able to opt for a final evaluation, the student should communicate that decision to the coordinator of the subject following the University regulations. If any of the proposed activities cannot be performed for whatever reason, equivalente alternative activities will be scheduled and published EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT Firts call rules apply for the second call. MANDATORY MATERIALS Micrcontroller based PCB and related IDEs and compilers. BIBLIOGRAPHY **Basic bibliography** [01] Microcontroladores MCS-51. Apuntes de clase de Ángel Mª Aledo Amorós [02] Prácticas Básicas con microcontroladores. Apuntes de José Miguel Gil-García Detailed bibliography Microcontroladores MCS-51 y MCS-251. J. Matas y R.R. Ramos. Edicions UPC. 2001 - C and the 8051 Vol.I y II. Thomas W.Schultz Journals r Web sites of interest In every chapter of [02] interesting URLs will be provided

In every chapter of [02] interesting URLs will be provided www.embedded.com www.8052.com

OBSERVATIONS

In the evaluation tests, only non-programmable scientific calculators are allowed to be used. If the device is programmable, the calculator will be retired and no additional device will be allowed, even if it fulfills the requirements. In case cheating is detected, the protocol about academic ethic issued by the University of the Basque Country will be followed.

COURSE GI	JIDE	2023/24		
Faculty	163 - Faculty o	f Engineering - Vitoria-Gasteiz	Cycle	•
Degree	GIEIAU10 - Ba	chelor's Degree in Industrial Electronics and Automation Engine	Year	Fourth year
COURSE				
26005 - E	mbedded System	S	Cr	edits, ECTS: 6
COURSE DE	ESCRIPTION			
subjects s nowadays	studying current m s. It follows the 3th	year optional subject that pursues learning a set of modern tools icrocontroller architecture and tools employed in developing micr year starting subject about microcontrollers (Digital Electronic S cks. It complements the subject Industrial Informatics.	ocontroll	er based systems
COMPETEN	CIES/LEARNING	RESULTS FOR THE SUBJECT		
C3 ̵ versatility C4 ̵ transmit k C6 ̵ C10 R TEEOI3 & As outcor implemen must be p manufact	1; Knowledge on to adapt themselv 1; Capability to so nowledge and ski 1; Capability to ha 211; Capability to ha 211; Capability to ha 211; Knowledge ne of the developer tation problems wo present. They will a	s from the degree verified report will be developed: basic and technologic topics, which will allow learning new metho ves to new situations. olve problems with initiative, decision-making, creativity, critic thin lls in engineering fields. andle specifications, regulations and rules of mandatory complian work in a multidisciplinary and multilingual environment ge of the foundation and applications of electronic circuits and mic ment of the aforementioned competencies, the student will be abl ith microcontrollers including several peripherals and communica- also have to introduce publicly the features of selected microcont	iking and nce. crocontro e to solv ations sta	to convey and llers. e basic cks where TCP/IP
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	-			
Master cla as guiding	asses will use orig g thread to introdu	inal datasheets, manuals, user's guides and manufacture ce concepts and develop competencies. It is pursued that studer development tools. The demonstrations will be coordinated with	nts have	direct contact with

Universidad Euskal Herriko del Pais Vasco Unibertsitatea

	Types of teaching	М	S	GA	GL	GO	GCL	ТА	TI	GCA]
	Hours of face-to-face teaching	30			30		002				
Horas de Activ	vidad No Presencial del Alumno/a	45			45						
Legend:	M: Lecture-based GL: Applied laboratory-based grou TA: Workshop	ps GC	• •	d compu al worksł		d groups	GCL:	Applied	clinical-l	n-based g based gro k groups	oups
Evaluation m	ethods										
- End-of-co	ourse evaluation										
valuation to	ols and percentages of final	mark									
- Exercises	s, cases or problem sets 10% l assignments 90%										
	XAMINATION PERIOD: GUID	ELINE	ES ANI		NG OU	Т					
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OBSERVATIONS

NAZIOARTEKO BIKAINTASUN CAMPUSA CAMPUS DE EXCELENCIA INTERNACIONA

sidad Euskal Herriko Unibertsitatea

In the evaluation tests, only non-programmable scientific calculators are allowed to be used. If the device is programmable the calculator will be retired and no additional device will be allowed, even if it fulfills the requirements. In case cheating is detected, the protocol about academic ethic issued by the University of the Basque Country will be followed.

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COURSE GUI	DE	2023/24									
Faculty	163 - Faculty of	of Engineering -	Vitoria-Gast	eiz					Cycle	e	•
Degree	GIEIAU10 - Ba	achelor's Degree	e in Industria	l Electro	nics ar	nd Auto	mation	Engine	Year		Fourth ye
COURSE											
26006 - Ext	tended Industria	al Information Te	echnology							Credits	s, ECTS:
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- Written test, open questions 20%

- Exercises, cases or problem sets 40%
- Teamwork assignments (problem solving, Project design) 40%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

- Own Theoretical Work: 40%
- Practical work: 40%
- Tests on other theoretical and / or practical work: 20%

According to the Regulations governing the Evaluation of Students in official Bachelor's degrees, chapter II, article 8, section 3, all students will have the right to be evaluated through the final evaluation system, regardless of whether or not they have participated in the system continuous evaluation. To do this, students must submit in writing to the teaching staff responsible for the subject the waiver of continuous assessment. The students will have a period of 9 weeks from the beginning of the subject, in accordance with the academic calendar of the center.

If a student wishes to waive the Ordinary Call, he may do so by writing to the teaching staff of the subject in the dates designed by the normative.

If there is a final written test, a minimum qualification of 4/10 must be obtained in order to pass the subject.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

The same criteria will be used as in the Ordinary Call. In this sense, the necessary tests will be carried out to ensure an adequate evaluation of the subject.

If applicable, the qualifications corresponding to the continuous evaluation of the subject may be kept.

MANDATORY MATERIALS

The teaching materials will be made available to through the e-gela platform or the teacher's website.

BIBLIOGRAPHY

Basic bibliography

The manuals and examples provided by the manufacturers of the automata will be used.

Detailed bibliography

If necessary, additional in-depth bibliography as well as on-line references will be provided during the development of the subject.

Journals

If necessary, additional in-depth bibliography as well as on-line references will be provided during the development of the subject.

Web sites of interest

If necessary, additional in-depth bibliography as well as on-line references will be provided during the development of the subject.

OBSERVATIONS

COURSE GUIDE	2023/24			
Faculty 163 - Faculty of	of Engineering - Vitoria-Gasteiz	Cycle].	
Degree GIEIAU10 - Ba	achelor's Degree in Industrial Electronics and Automation Engine	'ear	Fourth yea	ar
COURSE			_	
26007 - Intelligent Control		Credi	ts, ECTS:	6
COURSE DESCRIPTION				
engineering. Consequently discrete systems and signal discretize existing continue SUBJECT DESCRIPTORS Control Theory and Feedb Dynamics and Stability; Cy PREVIOUS REQUIREMENT This subject is aimed at str Control and (2) Computer Some students may come required to have previous of programming algorithms we students coming from the for control theory but are less students will produce multi COMPETENCIES/LEARNING Computer control will deve BASIC COMPETENCES 1. Acquire knowledge about 2. Learn new methods and 3. Solve problems with init 4. Communicate and trans 5. Working in a multilingua 6. Apply the scientific meth a) Analyze problems and b) Make hypothesis c) Find solutions 7. Working both autonomo LEARNING OUTCOME 1. Learning the role of the 2. Learning the role of the	Ack; Computer role in control systems; Signals and Systems; Discre- ber-Physical Systems (CPS) NTS Jidents from two different engineering degrees, namely (1) Industrial Management and Information Systems. from the degree at Computer Management and Information System chowledge in Control Theory. These students have a sound backgr hich are basic tools for implementing complex computer control system and are basic tools for implementing complex computer control system industrial Electronics and Automatic Control degree are already farr ss familiar with computers and programming tools. The combination disciplinary working groups which is a basic learning competence. B RESULTS FOR THE SUBJECT lop different competences and producing the following learning out at basic and technological matters versatility to adapt to new situations ative and creativity, providing innovative solutions mit knowledge in the Electronics & Computing domains I and multidomain environments. todology strategies: I situations usly and in groups computers in modern control systems ealing with signals	ntrol system ntrol based ete system I Electronio ns, conseq ound in co stems. On niliar with th n of differe	ms; (3) mod l systems a ls; Modeling cs and Auto uently they mputers an the contrary he basic co	nd (6) g; omatic are no
 Ability of modeling discrete Capability for analyzing 	the stability of systems simple computer based systems			
Theoretical and Practical Co	ontents			
CONTENTS: 1. Basic introduction to cor 2. The role of the compute 3. Discrete signals: a) Mathematical represe b) Signal sampling and re 4. Discrete systems:	r in control			

- a) Modelling of discrete systems (Difference equations / The Z transform / Block diagrams)
- b) Transient vs. steady state response analysis

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- c) Relationship between Laplace and Z transforms
- d) Composing complex systems Algebra of blocks
- 5. Stability analysis of discrete systems
- 6. Design of discrete control systems
 - a) Discretisation of continuous controllers
 - b) Digital Direct Control

NOTE:

These topics will be developed both in the classroom and in the laboratory.

TEACHING METHODS

CLASSROOM

1. Lectures will be used to explain the major concepts of this module.

2. Some collaborative activities will be included and the deliverables will be required. These deliverables will be used in the qualification of the students.

LABORATORY

- 1. Some exercises will be proposed to be solved by the students.
- 2. The students will have to create a short project where they must apply the concepts learnt during this module.

TYPES OF TEACHING

	Types of teaching	М	S	GA	GL	GO	GCL	ТА	TI	GCA	
	Hours of face-to-face teaching	30			30						
loras de Activ	pras de Actividad No Presencial del Alumno/				45						
Legend:	M: Lecture-based			GA: Applied classroom-based group							
	GL: Applied laboratory-based grou	ps GO	D: Applie	d compu	ter-base	GCL: Applied clinical-based groups					
	TA: Workshop	TI:	TI: Industrial workshop					GCA: Applied fieldwork groups			

Evaluation methods

- Continuous evaluation
- End-of-course evaluation

Evaluation tools and percentages of final mark

- Individual assignments 10%
- Teamwork assignments (problem solving, Project design) 30%
- Oral presentation of assigned tasks, Reading; 10%
- Portfolio 50%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

A. CONTINUOUS ASSESMENT of the tasks:

1. Portfolio of the tasks proposed during the laboratory sessions (30%)

- Reports of the Matlab exercises proposed in the lab.
- 2. Realization of a proposed miniproject related with the concepts learnt in the classroom (50%)
- 3. Presentation of the miniproject (10%)
- 4. Realization of other individual tasks (10%)

NOTES FOR THE CONTINUOUS ASSESMENT:

Students will be qualified as NON-PRESENTED when they do not completed the amount of the 55% of the total final assignments.

The deliverables associated to these tasks SHOULD be submitted by means of eGela.

B. ALTERNATIVE FINAL EXAM (100%)

The alternative final exam may include two parts, one written part with some questions and problems related to the syllabus of the module and another part in the laboratory where the students should solve some exercises with Matlab.

C. RENOUNCE PROCEDURE

Students will automatically renounce to the module when they do not present in time the 55 % of the proposed tasks of the module. In case these students want to pass the module, they should do the alternative final exam.

All students may renounce to present to the final exam by communicating it to the lecturer at least 15 days in advance.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

EXTRAORDINARY FINAL EXAM (100%)

The extraordinary final exam may include two parts, one written part with some questions and problems related to the syllabus of the module and another part in the laboratory where the students should solve some exercises with Matlab.

All students may renounce to present to the final exam by communicating it to the lecturer at least 15 days in advance.

MANDATORY MATERIALS

Matlab, Arduino

BIBLIOGRAPHY

Basic bibliography

- 1. K. Ogata, Discrete-Time Control Systems, 2nd Ed., Prentice-Hall, 1995.
- 2. R.C. Dorf, R.H. Bishop, Modern Control Systems 12th Ed., Pearson, 2010.
- 3. eGela: [https://egela.ehu.es/]
- 4. Tutorial on Matlab and control: [http://ctms.engin.umich.edu/CTMS/]

Detailed bibliography

- 1. C.L. Philips, H.T. Nagle, Digital control systems. Analysis and Design, Prentice-Hall, 1995
- 2. R. G. Jacquot, Modern Digital Control Systems, Marcel Dekker, Inc., 1984.

3. K.J. Astrom, R.M. Murray, Feedback Systems, Princeton University Press, 2009.

4. K.J. Astrom and B. Wittenmark, Computer Controlled Systems, Prentice Hall. 1992

5. Franklin, G. F., Powel, J.D. and Workman M.L., Digital control of Dynamic Systems, 2nd Ed. Addison Wesley.1992

6. E.A. Lee, S.A. Seshia, Introduction to Embedded Systems. A Cyber-Physical Approach, UC Berkeley, 2012. Available at: [http://leeseshia.org/releases/LeeSeshia_DigitalV1_50.pdf]

Journals

Automática (Elsevier) International Journal of Control Control System Magazine (IEEE)

Web sites of interest

1. eGela: [https://egela.ehu.es/]

2. Tutorial on Matlab and control: [http://ctms.engin.umich.edu/CTMS/]

OBSERVATIONS

COURSE GUIDE 2023/24							
Faculty 163 - Faculty of Engineering - Vitoria-Gasteiz	Cycle].					
Degree GIIGSI10 - Bachelor's Degree in Computer Engineering in Management and In Year First year							
COURSE							
26013 - Methodology of Programming	Credi	its, ECTS:	6				
COURSE DESCRIPTION							

COURSE DESCRIPTION

The subject of Programming Methodology introduces the student to the field of formal software development. In this subject, which is taught after Basic Programming, the basic concepts necessary to create and execute computer programs will be introduced, emphasizing formal specification methods.

COMPETENCIES/LEARNING RESULTS FOR THE SUBJECT

- Capacity to analyse, design, construct and maintain applications in a robust and secure manner
- Capacity to specify, document, validate and verify programs
- Capacity to reason and justify properties related to programs
- Capacity to evaluate and compare specifications and documentation of programs from the quality point of view
- Knowledge of formal specification and program design methods
- Knowledge of basic notions of axiomatic semantics of programming languages
- Knowledge and application of basic algorithmic procedures of IT technologies for the design of solutions to problems, analysing the suitability and complexity of the algorithms proposed

Specific objectives:

- Specification, correction and design by contract
- Formal pre-post specification
- Hoare's formal system
- The equational specification technique
- The formal derivation method

CONTENIDOS TEÓRICO-PRÁCTICOS

- Topic 0: Introduction
- Topic 1: Logical Assertions
- Topic 2: Specification
- Topic 3: Verification
- Topic 4: Derivation

TEACHING METHODS

In the lectures there will be sessions of concepts exposition, reinforced with examples of situations in which these concepts are to be used.

In the practical part a set of exercises will be developed. The exercises to be carried out pose programming problems that students must perform in the most autonomous way as possible.

During the lectures group work will be carried out systematically, discussing and presenting the results of exercises with the aim of encouraging direct participation in the course and also students' motivation.

TYPES OF TEACHING

	Types of teaching	М	S	GA	GL	GO	GCL	ТА	TI	GCA		
	Hours of face-to-face teaching	45		15								
Horas de Activ	vidad No Presencial del Alumno/a	67,5		22,5								
Legend:	M: Lecture-based	S:	Seminar				GA: A	pplied classroom-based groups				
	GL: Applied laboratory-based grou	ps GC	D: Applie	d compu	ter-base	d groups	GCL:	GCL: Applied clinical-based groups				
	TA: Workshop	TI:	Industria	al worksh	юр		GCA: Applied fieldwork groups					
Evaluation m	ethods											
- Continuo	us evaluation											
- End-of-co	ourse evaluation											
valuation to	ols and percentages of final	mark										

- Written test, open questions 100%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

Assessment of the subject in the ordinary call is either done through continuous assessment or final assessment. By default, all students will do continuous assessment, unless they withdraw from it.

* Evaluation with final exam

It is allowed to withdraw from continuous assessment to take a single final exam. This withdrawal must be communicated the professor responsible for the course in the period set out in Article 8 of the Rules for Student Assessment. Withdrawals from continuous assessment will not be accepted after the dates stated, except in exceptional cases.

The exam will test the entire content of the subject. To pass the subject, it is necessary to obtain a minimum mark of 5 out of 10 in the exam.

* Continuous assessment

By default, continuous assessment is the standard method, unless the student has expressly withdrawn from it.

Continuous assessment is done through written tests to demonstrate that knowledge and competences taught in the subject have been acquired.

1) First-order logic (20% of the grade)

2) Sequences and conditionals (20% of the grade)

3) Iterations (30% of the grade)

- 4) Recursiveness and function calls (30% of the grade)
- * Requirements to pass the ordinary call through continuous evaluation without final exam
- Perform all written tests
- Minimum score of 5 in the sum of all the tests
- Minimum score of 25% on each of the tests
- * Ordinary final exam:

The only students who can present themselves are those who have passed at least three written tests, although they may not have reached the level required to pass an ordinary call in continuous assessment.

The exam will be divided into four parts, corresponding to each test:

The student may take the parts he/she wishes

The grades from previous tests will be maintained if they are not taken in this exam

The grade of previously taken tests will not be maintained (corresponding to the parts taken in this exam) for the purposes of the calculation of the final grade

* Requirements to pass the ordinary call:

- Have taken all the written tests (in the ordinary final exam or before)
- Minimum mark of 5 in the sum of all the written tests
- Minimum mark of 25% in each written test

* Withdrawal from the exam:

Students following continuous assessment may withdraw from the call as per that stated in Article 12 of the Rules for Student Assessment

* Cases of copying:

Article 11 of the Rules for Student Assessment will be applied.

* Continuous assessment

By default, continuous assessment is the standard method, unless the student has expressly withdrawn from it. Continuous assessment is done through written tests, in which students must demonstrate that they have acquired the knowledge and competences taught in the subject.

- 1) First-order logic (20% of the grade)
- 2) Sequences and conditionals (20% of the grade)
- 3) Iterations (30% of the grade)
- 4) Recursiveness and function calls (30% of the grade)
- * Requirements to pass the ordinary call through continuous evaluation without final exam
- Perform all written tests
- Minimum score of 5 in the sum of all the tests
- Minimum score of 25% on each of the tests

* Ordinary final exam:

The only students who can present themselves are those who have passed at least three written tests, although they may not have reached the level required to pass an ordinary call in continuous assessment.

The exam will be divided into four parts, corresponding to each test:

The student may take the parts he/she wishes

The grades from previous tests will be maintained if they are not taken in this exam

The grade of previously taken tests will not be maintained (corresponding to the parts taken in this exam) for the purposes of the calculation of the final grade

* Requirements to pass the ordinary call:

Have taken all the written tests (in the ordinary final exam or before) Minimum mark of 5 in the sum of all the written tests Minimum mark of 25% in each written test

* Withdrawal from the exam:

Students following continuous assessment may withdraw from the call as per that stated in Article 12 of the Rules for Student Assessment

* Cases of copying:

Article 11 of the Rules for Student Assessment will be applied.

* If the course cannot be assessed face-to-face, the relevant changes will be made to carry it out online by using the IT tools available at the UPV/EHU. The particulars of this online assessment will be made public.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

The extraordinary call consists of an exam on 10 points in which the whole of the subject is evaluated.

To pass the subject, it is necessary to obtain a minimum score of 5 points out of the total of 10 exam points.

To waive the right to the examination, it will be sufficient not to appear. In this case, the qualification will be "No Presented".

* Note that no grade from the ordinary call is kept

* Cases of copying

Article 11 of the Rules for Student Assessment will be applied.

MANDATORY MATERIALS

- Material of the subject (Notes and Laboratories)
- eGela virtual platform of the UPV/EHU for this subject

BIBLIOGRAFÍA

Basic bibliography

- Especificación, Verificación y Derivación Formal de Programas. J. Álvez, X. Arregi, J. Gaintzarain, P. Lucio y M. Marichalar, Pearson, 2016.

- Programen Egiaztapena eta Eratorpena. X. Arregi, A. Díaz de Ilaraza, P. Lucio. Udako euskal Unibersitatea, 1993.

- Metodología y Tecnología de la Programación II. M. Díaz Roca, J. C. Rodríguez del Pino. Univ. Palmas Gran Canaria, 2004.

- Verificación de Programas y Metodología de la Programación. A. Díaz de Ilarraza, P. Lucio. Servicio Editorial Universidad del País Vasco, 1990.

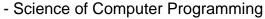
- Program Derivation. The Development of Programs from Specifications. G. Dromey. Addison-Wesley, 1989.

Detailed bibliography

- Program Construction and Verification. R. C. Backhouse. Prentice-Hall, 1986.
- Programación metódica. J.L. Balcázar. MacGraw-Hill, 1993.
- Diseño de Programas.- Formalismo y Abstracción. Ricardo Peña. Prentice Hall, 1998.

Journals

- Acta Informatica
- Programming and Computer Software



- Science of Computer Programming
 Software Quality Journal
 Transactions on Software Engineering and Methodology

Web sites of interest

t,)

- http://www.sc.ehu.es/jiwlucap/metodologia.html
 http://en.wikipedia.org/wiki/Formal_verification
- http://en.wikipedia.org/wiki/Design_by_contract

OBSERVATIONS

COURSE GUI	DE	2023/24					
Faculty	163 - Faculty of	f Engineering - Vitoria-Gasteiz	Cycle].			
Degree	GIIGSI10 - Bac	Year	Year Second year				
COURSE							
26017 - So	ftware Engineeri	ng	Cred	its, ECTS:	6		
COURSE DES	SCRIPTION						
previously software ar To be able	captured. To dev chitectures will b to study Softwar	e course is to design and implement applications, the requireme velop software products following a systematic process, active n be applied, relying on tools that improve the quality of the software re Engineering without undue difficulty, it is recommended to have	nethodologie re.	s and multi-			
* Know and elements * Understa inheritand * Develop * Knowledg programs - In the sub * Knowledg Stacks, G * Knowledg Search, S * Ability to	odular and Object d understand the in an Object Ori- and the difference ce and polymorph small programs a ge and use of ex bject "Data Struct ge and ability to a Queues, Hash Ta ge and ability to a Sorting, and Enur efficiently select	applying all the concepts about programming acquired ceptions as an error control mechanism for the correct operatio cures and Algorithms" subject: apply Abstract Data Types to problems of medium complexity: L bles, Trees and Graphs analyze the main algorithms for the treatment of data structures	s, n of Lists, : blem				
C3: Know h C4: Know h Transversa	now to design a s now to implemen Il competencies:	nd an object-oriented software system in the UML language. software system in a multi-layer architecture based on the analy t a system based on the design of the application. icate and transmit knowledge, abilities and skills of the professi		-			
CONTENIDO	S TEÓRICO-PR	ÁCTICOS					
- Motivatio - Objective UT2: Speci - Study of UT3: Multi- - Design o UT4: Objec - Function UT5: Imple	fication of UML a the different artif layer software ar of the different lay ct Oriented Design ality design mentation of a sp	of the software ad associated programming technologies. artifacts facts existing in UML rchitectures: Presentation, Business Logic and Data yers that make up a software system in and Programming					
TEACHING M	ETHODS						
necessary concepts a The COMP	in the practical cl cquired through t PUTER PRACTIC	ical subject, the MASTER classes (M) will be used for the exposi lasses, as well as for the resolution of doubts raised by the stud the resolution of exercises will be reinforced, either individually CES (CP) classes will be used to apply the active Project-Based students will be provided with a project statement that realistica	ents. In the s or in small gr Learning me	same way, tl oups. ethodology.	he At the		
of the subje methodolog	ect. They will car gy proposes to de	ry out this project in groups of 2-4 people, following the agile SC evelop the project in an incremental way, through successive ite hat adds new functionality to the previous one. Each iteration is	RUM methor erations, in e	dology. This ach of which	s na		
		Páge : 1 / 4					

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realization of	of its correspon	ding documen	tation.

Legend: M: Lecture-based S: Seminar GA: Applied classroom-based groups GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-based groups TA: Workshop TI: Industrial workshop GCA: Applied fieldwork groups Valuation methods - - - Continuous evaluation - - - End-of-course evaluation - - valuation tools and percentages of final mark - - Written test, open questions 60% - Teamwork assignments (problem solving, Project design) 40% RDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT The evaluation of the subject in the ORDINARY call will be made through continuous evaluation or final evaluat default, all students will take continuous assessment, unless they resign it.		Types of teaching	М	S	GA	GL	GO	GCL	ТА	TI	GCA	
Legend: M: Lecture-based S: Seminar GA: Applied classroom-based groups GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-based groups TA: Workshop TI: Industrial workshop GCA: Applied fieldwork groups Evaluation methods - - - Continuous evaluation - - - End-of-course evaluation - - valuation tools and percentages of final mark - - - Written test, open questions 60% - - Teamwork assignments (problem solving, Project design) 40% RDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT The evaluation of the subject in the ORDINARY call will be made through continuous evaluation or final evaluat default, all students will take continuous assessment, unless they resign it.		Hours of face-to-face teaching	45			15						
GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-based groups TA: Workshop TI: Industrial workshop GCA: Applied fieldwork groups valuation methods - - - Continuous evaluation - End-of-course evaluation valuation tools and percentages of final mark - - - Written test, open questions 60% - - Teamwork assignments (problem solving, Project design) 40% RDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT - The evaluation of the subject in the ORDINARY call will be made through continuous evaluation or final evaluat default, all students will take continuous assessment, unless they resign it.	Horas de Activ	vidad No Presencial del Alumno/a	67,5			22,5						
TA: Workshop TI: Industrial workshop GCA: Applied fieldwork groups Evaluation methods - <	Legend:	M: Lecture-based	S:	Seminar				GA: A	pplied c	lassroon	n-based g	roups
 Continuous evaluation Continuous evaluation End-of-course evaluation valuation tools and percentages of final mark Written test, open questions 60% Teamwork assignments (problem solving, Project design) 40% RDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT The evaluation of the subject in the ORDINARY call will be made through continuous evaluation or final evaluat default, all students will take continuous assessment, unless they resign it. 		GL: Applied laboratory-based grou	ips GC): Applie	d compu	ter-base	d groups	GCL:	Applied	clinical-l	based gro	ups
 End-of-course evaluation Evaluation tools and percentages of final mark Written test, open questions 60% Teamwork assignments (problem solving, Project design) 40% PRDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT The evaluation of the subject in the ORDINARY call will be made through continuous evaluation or final evaluat default, all students will take continuous assessment, unless they resign it. 		TA: Workshop	TI:	Industria	al worksh	пор		GCA:	Applied	fieldwor	k groups	
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 Teamwork assignments (problem solving, Project design) 40% RDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT The evaluation of the subject in the ORDINARY call will be made through continuous evaluation or final evaluat default, all students will take continuous assessment, unless they resign it. 												
ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT The evaluation of the subject in the ORDINARY call will be made through continuous evaluation or final evaluat default, all students will take continuous assessment, unless they resign it.		· · ·	ng, Proj	ject des	sign) 4	10%						
The evaluation of the subject in the ORDINARY call will be made through continuous evaluation or final evaluat default, all students will take continuous assessment, unless they resign it.			DELINE	ES AND		NG OU	т					
default, all students will take continuous assessment, unless they resign it.				-				ntinuou	s evalı	ation o	or final e	valuation
		-					•		e eraie			, and all of the
						•	•					
* EVALUATION THROUGH FINAL EVALUATION	* EVALUAT	TION THROUGH FINAL EVAL	UATIO	N								
	The studen	t can renounce the continuous	evalua	ation to	carry c	out the t	final ev	aluatior) This	resiana	ation mu	et he cub

The student can renounce the continuous evaluation to carry out the final evaluation. This resignation must be submitted in writing to the responsible teachers within the terms stipulated in Article 8.3 of the regulations on student assessment. Exceptional cases or waivers of continuous evaluation will not be accepted after the dates enabled for it.

Students who have waived continuous assessment will undergo a test that represents 100% of the course grade in which both theoretical and practical (competences addressed in the project) aspects worked on in the subject will be evaluated. To pass, the student must obtain a minimum grade of 5 out of 10 in

both parts (theoretical and practical). When a student fails the subject having passed the theoretical part or the practical part, that part will be kept for the extraordinary call.

The student can give up the continuous evaluation to carry out the final evaluation. This waiver will must submit in writing to the responsible faculty within the deadlines stipulated in Article 8.3 of the regulations about student assessment. Exceptional cases or waivers of continuous evaluation will not be accepted. after the dates enabled for it.

* CONTINUOUS ASSESSMENT

The default assessment of the subject is done through continuous assessment, unless the student has expressly waived it.

The final grade for the course is calculated based on the test scores and a project to be developed as a team. The final grade is calculated with the following weights:

- 40% based on a series of deliveries on a project (deliverables).

- 60% by taking different tests that will always include 3 exams.

In this evaluation, the student takes two exams throughout the semester.

The student must obtain at least 4 out of 10 in each of the exams, being the average at least 5 out of 10, and 5 out of 10 in the project in order to pass. Otherwise, the grade obtained will be the average of the exams (and in no case will it exceed 4 points out of 10). When a student fails the subject having

passed the theoretical part (has obtained an average mark in the exams equal to or greater than 5, obtaining at least 4 points out of 10 in each of the partials) or the practical part, that part will be kept for the extraordincary call.

If it is not possible to carry out a face-to-face evaluation of the subject, the pertinent changes will be made to do it online by using the existing computer tools at the UPV / EHU. The characteristics of this online evaluation will be made public.

* WAIVER OF THE RIGHT TO EXAM

The student or the student who, having selected to take the final evaluation, does not appear for the exam in the ordinary call, will obtain the final grade "Not Presented".

The students who follow the continuous evaluation may make the waiver of the call in accordance with the provisions of Article 12 of the regulations on student evaluation.

* COPY CASES:

Article 11 of the current regulations regarding the evaluation of students will apply.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

Students who have not passed the subject in the ordinary call will undergo a test that represents 100% of the the course mark in which both theoretical and practical (competences addressed in the project) aspects worked on in the subject will be evaluated. To pass, the student must obtain a minimum grade of 5 in both parts (theoretical and practical). The theoretical part will account for 60% of the final mark and the practical part 40%. Students who have passed one of the parts (theoretical or practical) in the ordinary call must only do the pending part.

To waive the right to the exam, it will be enough to not show up.

If it is not possible to carry out a face-to-face evaluation of the subject, the pertinent changes will be made to do it online by using the existing computer tools at the UPV / EHU. The characteristics of this online evaluation will be made public.

* COPY CASES:

Article 11 of the current regulations regarding the evaluation of students will apply.

MANDATORY MATERIALS

Material shared in eGela.

BIBLIOGRAFÍA

Basic bibliography

Ingeniería del Software. Un enfoque práctico. Roger S. Pressman. MacGraw-Hill, 2001. 5a Edición.

Ingeniería El Proceso Unificado de Desarrollo de Software Jacobson, Booch, Rumbaugh. Editorial Addison Wesley, 1999

Design Patterns, Elements of Reusable Object-Oriented Software. Erich Gamma, Richard Helm, Ralph Johnson, John M. Vlissides, 1995

A Pattern Language: Towns, Buildings, Construction. Christopher Alexander, Sara Ishikawa, Murray Silverstein, 1977

Java 8 in Action

Lambdas, streams, and functional-style programming. Raoul-Gabriel Urma, Mario Fusco, and Alan Mycroft. Manning. 2014

Detailed bibliography

Object Oriented Software Construction. Bertrand Meyer. Prentice-Hall. 1998.

Journals

Web sites of interest

http://www.uml.org/

http://www.visual-paradigm.com/product/vpuml/

Books about design patrones: http://hillside.net/patterns/books/ http://www.javacamp.org/designPattern/ http://www.dofactory.com/net/design-patterns



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OBSERVATIONS

The concepts covered in the subjects "Modular and Object Oriented Programming" and "Data Structures and Algorithms" are required for this subject. To take this subject, you should have passed or at least completed these subjects.

COURSE GUI	DE	2023/24											
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ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

At least 5 points out of 10 must be scored to pass the subject course. There will be three partial examinations, each accounting for 25% of the course mark. For the ordinary session there will

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be an optional catch-up session for each of the three blocks to boost marks.

Laboratory work will be worth 25% of the mark, with individual testing after completion.

If continuous assessment is waived there will be a final examination worth 100% of the mark. The lecturer must be informed of this waiver by the 9th week, in accordance with current regulations.

If the final is not attended it will be considered not sat.

If the course cannot be assessed face-to-face, the relevant changes will be made to carry it out online by using the IT tools available at the UPV/EHU. The particulars of this online assessment will be made public.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

For the ordinary session there will be a final examination worth 100% of the mark. No mark is carried over from previous assessments.

MANDATORY MATERIALS

Course notes and JFLAP simulator.

BIBLIOGRAFÍA

Basic bibliography

Ramón F. BRENA; "Autómatas y Lenguajes. Un enfoque de diseño", Tec de Monterrey, 2003.

Elaine RICH, "Automata, Computability and Complexity. Theory and Applications", Pearson/Prentice Hall, 2008

V. MATHIVET, "Inteligencia Artificial para desarrolladores. Conceptos e implementación en Java", ENI Ediciones, 2017

J.E. HOPCROFT, R. MOTWANI, J.D. ULLMAN: "Teoría de Autómatas, Lenguajes y Computación" 3ª ed. Pearson educación, 2007

S. RUSSELL, P. NORVIG: "Artificial Intelligence: A Modern Approach" 2ª ed. Prentice Hall, 2003

S.H. RODGER, T.W. FINLEY; "JFLAP: An Interactive Formal Languages and Automata Package". Jones and Bartlett, 2006

Detailed bibliography

S. ARORA, B. BARAK: "Computational Complexity: A Modern Approach" Cambridge University, 2009.

D. WOOD; "Theory of computation". John Wiley & Sons, 1987.

T. MITCHELL: "Machine Learning" McGraw Hill, 1997

G.F. LUGER, W.A. STUBBLEFIELD: "Artificial Intelligence. Structures and Strategies for Complex Problem Solving." Benjamin/Cummings Publishing Company, Inc, 1998.

Journals

Web sites of interest

Java Computability Tool kit (JCT): http://humboldt.sunyit.edu/jct/ Visual and interctive tools (JFLAP): http://www.jflap.org/ Machine Learning theory and examples: http://www.cs.cmu.edu/~avrim/ML07/index.html Implementación de algoritmos de IA en Java: https://github.com/aima-java/aima-java

OBSERVATIONS

Faculty	163 - Faculty c	of Engineering - Vitoria-Gasteiz	Cycle .	
Degree		chelor's Degree in Computer Engineering in		
COURSE				cond year
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Definition Chapter 2 Definition Chapter 3 Description Chapter 4 Dual form Chapter 5 Study of t Chapter 6 Problem 9 Chapter 7	2: Linear programs of a lineal progra 3: Simplex method on, uses, variation 4: Duality. nulation and prope 5: Analysis of sens the changes in the 6: Integer linear pr solving when the 7: Model of transp	m and graphic solution. d. as and computational issues. erties. sibility. e solutions due to changes in the initial condi rogramming. variables have to take only integer values. S	itions.	
FEACHING	METHODS			
During pa students;	art of the sessions therefore, particip al exercises. The n.	, the lecturer will explain the concepts about bative sessions will be greatly encouraged. C students will also work in groups developing	Complementarily, some sessions will	be dedicate art of the

All teaching material will be available in the virtual classroom and students will have at their disposal a wide range of virtual tools for studying and communicating with their colleagues and with the lecturer.

NOTE: In case that the lessons and/or exams could not be done face-to-face, alternative options will be set up in order to replace them by on-line activities, making use of the tools provided by the university. The characteristics of the on-line evaluation will be published in the virtual classroom and in an ammendement of the academic guide.

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L]
Legend: M: Lecture-based GL: Applied laboratory-based TA: Workshop	groups G		r ed compu al worksh		d groups	GCL:	Applied	clinical-l	n-based g based gro k groups	oups
Evaluation methods										
- End-of-course evaluation										
valuation tools and percentages of fir	nal mark									
 Written test, open questions 60% Teamwork assignments (problem so Actividades en el aula virtual 20% 	lving, Pro	oject de	sign) 2	20%						
RDINARY EXAMINATION PERIOD: G	UIDELIN	ES AN		NG OU	Т					
Final exam: 60% (a minimum result of Continuous evaluation by means of ex Work in group (20%).		,		the sen	nester (20%).				
Students who do not want to follow the with a final exam.	e aforeme	entionec	d evalua	ation sy	stem wi	ill have	the pos	ssibility	/ to be e	valuated only
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During the final exam, study notes can a) the material for consultation is for o b) only documents in print will be allo c) notes will never got mixed with the to check them at any moment but will l d) students cannot add to the exam a	each spe wed, no exam. D nave to le	cific stu electron uring th et them	dent an lic devic le exerc back to	d cann ces (suc cise, no the ver	ot be sh ch as la tes will ry place	nared. ptops, t be plac after c	ablets ed at th	or mot ne side	•	
Students will carry their calculators for improve the drawing up of charts.	the exan	n, as we	ell as so	me bas	sic mate	erials fo	r drawi	ng (e.ç	g. a rulei	r) in order to
NOTE: In case that the lessons and/or replace them by on-line activities by m ebaluation will be published in the virtu	eans of t	he tools	s provide	ed by th	ne unive	ersity. T	he cha	iracteri	stics of	
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By means of a written exam (100% of extraordinary call.	the mark). Previo	ous wor	k done	along t	he sem	ester w	vill not	be cons	idered in the
NOTE: In case that the lessons and/or replace them by on-line activities by m evaluation will be published in the virtu	eans of t	he tools	s provide	ed by th	ne unive	ersity. T	he cha	iracteri	stics of	

Universidad Euskal Herriko del Pais Vasco Unibertsitatea

BIBLIOGRAFÍA

£.)

Basic bibliography

"Operations research : an introduction" Taha, Hamdy A. Ed. Prentice-Hall

"Linear programming : foundations and extensions" Robert J Vanderbei Ed. Springer

Detailed bibliography

"Elementary linear programming with applications" Bernard Kolman, Robert E Beck, Robert Edward Beck San Diego Academic Press

Journals

https://link-springer-com.ehu.idm.oclc.org/journal/41274 https://www-sciencedirect-com.ehu.idm.oclc.org/journal/european-journal-of-operational-research

Web sites of interest

https://www.ehu.eus/es/web/dma

OBSERVATIONS

	163 - Faculty of	Engineering - V	/itoria-Gas	teiz					Сус	le		
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COURSE DE												
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Universidad Euskal Herriko del Pais Vasco Unibertsitatea

- Written test, open questions 60%
- Exercises, cases or problem sets 40%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

Obligatory tasks of different kinds will be set during the course, both individually and as a group. Deliverables will consist of reports and presentations to give in class, assessed by the lecturer and by colleagues according to predetermined criteria.

For continuous assessment practical work, reports and presentations will be worth 40% of the final mark. There will be a final written examination worth 60% of the final mark, in which a minimum of 3.5 out of 10 must be scored to pass the course.

If continuous assessment is waived there will be a final examination worth 100% of the mark. The lecturer must be informed of this waiver by the 9th week, in accordance with current regulations.

If the final examination is not attended it will be considered not sat.

If the course cannot be assessed face-to-face, the relevant changes will be made to carry it out online by using the IT tools available at the UPV/EHU. The particulars of this online assessment will be made public.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

Assessment of the extraordinary session will be by a single written examination covering the topics seen in the lectures and laboratory work, updating cases from one year to the next. Relevant material produced by students will be made available for reference to the rest of the eGela platform.

MANDATORY MATERIALS

Class notes, classroom and laboratory teaching support material. Data Protection Act, Information Society Services Act and Digital Signature Act.

BIBLIOGRAPHY

Basic bibliography

Álvaro Gómez Vieites, "Enciclopedia de la Seguridad Informática", Ra-Ma 2011 (2ª edición actualizada)

Jesús Costas Santos, "Seguridad y Alta Disponibilidad", Ra-Ma 2011.

Detailed bibliography

ACISSI, "Seguridad Informática – Ethical Hacking. Conocer el ataque para una mejor defensa", 2ª ed,ENI ediciones, 2013

C.J. Bennett y C.D. Raab, "The governance of privacy", MIT Press 2006

Beyond Fear. B. Schneier, "Beyond Fear: Thinking Sensibly About Security in an Uncertain World", Springer, 2006.

Emilio del Perso et al, "Nuevo Reglamento de Protección de Datos de Carácter Personal. Díaz de Santos, 2008.

Journals

Auditoría + Seguridad informática

IEEE Security & Privacy

Web sites of interest

https://www.incibe.es/ http://www.criptored.upm.es/ http://www.intypedia.com/ http://www.rediris.es/ http://www.avpd.euskadi.eus http://www.agpd.es/



NAZIOARTEKO BIKAINTASUN CAMPUSA CAMPUS DE EXCELENCIA

idad asco Unibertsitatea

Univ del Paí

Faculty	163 - Faculty o	of Engineering - Vitor	ria-Gasteiz	[Cycle	٦.
Degree]		Computer Engineering in	Management and In	•	
DURSE						First year
	asic Programming	n			Cred	its, ECTS: 6
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		RESULTS FOR TH quired in the subject				
operation - Ability to - Ability to them. Ab - Knowled problem- - Knowled	ns, iteration, sub- divide a problem design simple algorithm ility to write code ge, design and ef solving. ge of basic algori	programs. into logical parts the gorithms to solve pro in accordance with fficient use of the mo thmic procedures in	f programs that contain: at can be solved (progra oblems, implement then rules of good practice. ost suitable types and st IT technologies for des exity of the algorithms pr	ammed) independently n, test them and fine-tu tructures of data for ign solutions to		
ONTENIDO	S TEÓRICO-PR	ÁCTICOS				
(specific	apter introduces		ing on the methodology tion and test). It highligh			
	ntary concepts fo nout this chapter,		m the basics of algorithr	n design and program	ming.	
	ons and procedur hapter the stude		n and implement subpro	ograms.		
This cha			vi structures. In additior	n, the main algorithm		
	tructures					

In the laboratory, a series of (previously-distributed) exercises will be worked on. The sessions require prior preparation and the presentation of a report using the problem-solving methodology given. A project will also be created, in which the student will put the concepts worked on in the subject into practice.

NAZIOARTEKO BIKAINTASUN CAMPUSA CAMPUS DE EXCELENCIA INTERNACIONA

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PES OF TE	ACHING									
	Types of teaching	Μ	S	GA	GL	GO	GCL	TA	TI	GCA
	Hours of face-to-face teaching	30				30				
loras de Activ	idad No Presencial del Alumno/a	45				45				
Legend:	M: Lecture-based	S	: Seminar	,			GA: A	pplied cl	assroon	n-based gro
	GL: Applied laboratory-based group	os G	GO: Applied computer-based groups					GCL: Applied clinical-based grou		
	TA: Workshop	Т	I: Industria	al worksh	юр		GCA:	Applied	fieldwor	k groups

Evaluation methods

- End-of-course evaluation

Evaluation tools and percentages of final mark

- Written test, open questions 55%

- Exercises, cases or problem sets 45%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

Assessment in the ORDINARY call will be done under continuous assessment or final assessment. By defect, all students will do continuous assessment unless they withdraw from it.

* FINAL ASSESSMENT

The student may withdraw from continuous assessment to do final assessment (final exam). This withdrawal will be presented in writing to the professor responsible for the subject in the periods established in Article 8 of the Rules on Assessment of Students. Exceptional cases will not be accepted, nor can students withdraw from continuous assessment after the stated dates.

* CONTINUOUS ASSESSMENT

The final grade of the subject is calculated on the basis of the exam marks (55%), laboratory work (35%) and individual practical work (15%).

GRADING IN MINUTES:

Students who have not withdrawn from continuous assessment will be considered as presented for the ORDINARY call.

To pass the subject, the student must take all the exams and obtain a minimum final grade of 5 out of 10. The student must also score at least 3.5 out of 10 in the grade corresponding to the exams in which the practical and laboratory work is included. Otherwise, the grade obtained will the average marks of the exams.

* WITHDRAWAL FROM THE EXAM

A student who, having opted for final assessment, does not present him/herself for the ORDINARY call will be graded as "NOT PRESENTED". A student who takes continuous assessment may withdraw from the call as per stipulated in Article 12 of the Rules on Student Assessment.

* CASES OF COPYING

Article 11 of the current rules on student assessment will be applied.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

Assessment in the EXTRAORDINARY call will consist of three parts:

In the extraordinary call, the students may recover the parts corresponding to practical work and laboratory work by answering some specific questionnaires.

If they have passes all these parts, they may maintain the grade obtained in the ordinary call.

Students who have withdrawn from continuous assessment will have a single final exam in which the aspects worked on in class, the laboratories and practical sessions will be assessed.

To withdraw from this exam, it is sufficient not to appear on the day.

MANDATORY MATERIALS

Course material in egela

BIBLIOGRAFÍA

Basic bibliography

"Una Introducción a la programación. Un enfoque algorítmico". J.J.Garcia, F.J. Montoya, J.L Fernandez, M.J. Majado Thomson Ed. 2005 "Oinarrizko Programazioa. Ariketa-bilduma" Díaz de Ilarraza A., Sarasola K.

Udako Euskal Unibertsitatea, 1999.

Detailed bibliography

"La práctica de la programación". B.W. Kernighan, R. Pike. Prentice Hall. Ed, 2000. Code Complete: A Practical Handbook of Software Construction. Steve McConnell. Microsoft Press, 2004.

Journals

Web sites of interest

es.wikibooks.org/wiki/Fundamentos_de_programación

OBSERVATIONS

		2023/24		_			
Faculty	163 - Faculty of	f Engineering - Vitoria-Ga	steiz	C	Cycle		
Degree	GIIGSI10 - Bac	chelor's Degree in Compu	ter Engineering in Manageme	nt and In Y	ear	Fourth ye	ar
COURSE							
26036 - D	evelopment of Gr	aphic Systems			Crea	lits, ECTS:	4,5
	ESCRIPTION						
knowledg This subject learning the concepts Modular F Software. The subject will work for coordinate work.Durite autonomod Although	e in various areas act is an introduction of specific concept and techniques of Programming and act is organized are the different stage among several p ong the project wor ously, learning both the subject does r a first approach to	s of knowledge such as gr on to the demanding work ots and techniques of prog f programming and softwa Object Orientation, Data ound two larger group pro- s of a software project. D programmers using tools f rk, the students will be sup h from the correct answer not teach everything nece the programming of grap	nes, are one of the most comp aphic programming, physics, s d of video game that focuses gramming, the students will im are engineering acquired in the Structures and Algorithms and bjects than those undertaken in ue to the entity of these project for collaborative development, pervised by the professor but the sand from the mistakes. ssary to enter directly into a p hic applications, which is a wo e/video processing, architectu	sound, artifi on graphic p plement in e subjects o d Engineerir n previous s ts, students skill that is will take the rofessional ork area with	cial intell program a group p f Basic P ng of the subjects, s will face essentia design o video ga	igence, etc. ning.In addit project more rogramming in which stu- the need to I for the work lecisions me studio, it	ion to gener , dents d of
Students from the c -G001 - A -G005 - A engineerin making, a of Compu -G012 - K	pursuing this subj competencies of the bility to develop re- bility to conceive, ng methods as an utonomy and creater ter Engineer.	ne module to which it is a eal-time graphical applica develop and maintain con instrument for quality ass ativity. Ability to communic plication of basic human r	c skills listed below. These ha part. tions. mputer systems, services and suranceG009 - Ability to solv cate and transmit the knowled esources management, proje	application e problems ge, skills an ct organizat	s using s with initia d abilitie ion and p	oftware ative, decisic s of the profe planningCB	on ession 5 -
	ents have develop and Practical Co	-	g skills to undertake further st	udies with a	ı high de	gree of autor	nomy.
1. Introdu 1.1. Defin 1.2. Basic 2. Graphic 2.1.Graph 2.1.1. Prin 2.1.2. Pro 2.1.3. Tex 2.2. Grap 2.2.1. Pro 2.2.2. Lig 3. Applica 3.1. Introd 3.2. Grap 3.3. Contr 3.4. [Intro	ction to graphics s itions. concepts: applica structure of graph c programming int nic programming in nitive drawing. jection matrices for tures. hic programming i jection matrices for tures. hic sol and animation. duction to collisior	systems. ation, graphic engine, API hic applications. terfaces. In two dimensions. or two-dimensional drawir in three dimensions. or drawing in three dimens ; interface.	ng.	Э:			
TEACHING							
The meth	odology of the sub	piect will rely heavily on th	e two group projects students	will work o	n. Everv	week, theore	etical

Universidad Euskal Herriko del Pais Vasco Unibertsitatea

TYPES OF TEACHING S GL GO GCL TA ΤI GCA Types of teaching Μ GA Hours of face-to-face teaching 15 30 Horas de Actividad No Presencial del Alumno/a 22,5 45 Legend: M: Lecture-based S: Seminar GA: Applied classroom-based groups GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-based groups **TI: Industrial workshop** GCA: Applied fieldwork groups TA: Workshop **Evaluation methods** - Continuous evaluation End-of-course evaluation Evaluation tools and percentages of final mark - Multiple choice test 20% - Teamwork assignments (problem solving, Project design) 80% **ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT** The assessment shall be carried out by continuous assessment. During the course, students will work on two group projects. The grade of the first project will account for 30% of the final grade, the second project 50%, and the remaining 20% will be evaluated by individual written tests. The two projects will be evaluated in the same way: -Group grade (50%) -Individual grade (20%) -Individual presentation (20%) -Cross evaluation (10%) The student may waive the continuous assessment and request the evaluation by final examination. In this case, subject to compliance with the deadlines laid down by the regulations, the assessment shall be carried out by means of a written examination comprising 100% of the student's note. **EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT** In the special call, the student will perform a written exam which will value 100% of the student's note. This examination will consist of test questions and practical exercises to be resolved. MANDATORY MATERIALS All necessary materials will be available in Egela BIBLIOGRAPHY **Basic bibliography** -Fundamentals of Computer Graphics 3rd Edition, Roger S. Pressman. A K Peters/CRC Press 2009, ISBN 978-1568814698 -Game Engine Architecture, Second Edition, Jason Gregory. A K Petres/CRC Press 2014. ISBN 978-1466560017 -Programming: Principles and Practice Using C++ (Second edition), Bjarne Stroustrup. Addison-Wesley 2014, ISBN 978-0-321-99278-9 **Detailed bibliography** -Software engineering. A Practitioner's Approach, Roger S. Pressman. McGraw Hill 2001, ISBN 84-81-3214-9 -Game Physics 2nd Edition, David H. Eberly. CRC Press 2010, ISBN 978-0123749031 Journals Web sites of interest http://unveopoper.nvidia.com http://www.gamedev.net/ **OBSERVATIONS**

COURSE GL		2023/24			-	
Faculty	163 - Faculty	of Engineering - Vite	oria-Gasteiz		Cycle	
Degree	GMECAN10 -	Bachelor`s Degree	in Mechanical Engin	eering	Year	Third year
COURSE						
26045 - E	lasticity and Stre	ngth of Materials			Crec	dits, ECTS: 9
COURSE DE	SCRIPTION					
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the body's simple ge accurate a	response. The sometry, and allow as those of the el CIES/LEARNING	dies elastic bodies, strength of materials w the use of simplify asticity theory, but t G RESULTS FOR T	formulating mathema s, studies the most co ring hypotheses that s the error can be cons	tically the relations mmon elements of speed up the calcu idered negligible.	of structures. Thes	se elements hav s are not as
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the body's simple ger accurate a COMPETEN The behave the Theory subjected pure bend structures - The subjected pure bend structures - The subjected - Learning - Know, un enable the areas of m - Properly solid: anal mechanics	response. The sometry, and allow as those of the electronic of the	dies elastic bodies, strength of materials w the use of simplify asticity theory, but the GRESULTS FOR T mable) solids is intre program focuses of on stresses. It starts bending are studied wledge that is at the lule Competence, Mapply the fundamental subsequent applical eering and also pro- gies of scientific me qualitatively and qualitatively qualita	formulating mathema s, studies the most co ring hypotheses that s the error can be cons THE SUBJECT roduced in this course on the analysis and d s with axial stress. Ne ed, and their applicati e base of the analysis lechanics:	tically the relations mmon elements of speed up the calcu- idered negligible. After an exposition esign of prismatic p ext, the stresses and on is carried out for and design methon sistance of materia hods and theories it versatility to adapt lems posed by stru- hypotheses and so	on of the fundame piece-type structu nd deformations of the resolution of ods of any Mecha rials to the behavior in their professior pt to new situatior uctural systems an olutions to solve p	e elements have s are not as ental concepts of ural elements, iriginated both in f isostatic nical Engineerin or of real solids the nal development ns. nd the deformation problems inhere

the development of own work, the elasticity and resistance of materials.

- Carry out measurements, calculations, studies, reports and other similar work related to problematic situations that may arise in the field of elasticity and resistance of materials.

Theoretical and Practical Contents

The elastic solid: stresses, deformations and compatibility equations.

- Tensionn and compression.
- Shear strength
- Flexure theory: pure, simple, compound, isostatic and hyperstatic.
- Torsion.
- Internal potential. Energy theorems

TEACHING METHODS

In the theoretical classes the theory will be explained and related examples will be solved.

Some topics will be worked on with the flipped classroom methodology, and material will be made available to the students to work on the theory at home, and doubts will be answered in class and exercises to apply the theory will be carried out.

In classroom practices, theoretical concepts can be explained and exercises to be developed proposed.

In class, the teacher will propose some work, which can be problems, practices or exercises to work on theory. All these works will be evaluated and will account for 20% of the final grade. During the semester, there will be a partial exam, which, if approved, will release material for the final exam.

To pass the exams, whether partial or final, you must obtain a minimum score of 3 out of 10 in each section of the same. Therefore, the final grade will be calculated as follows: 0.4 x partial exam grade + 0.4 final exam grade + 0.2 individual work grade.

TYPES OF TEACHING

Types of teaching	М	S	GA	GL	GO	GCL	TA	TI	GCA
Hours of face-to-face teaching	45		30	15					
Horas de Actividad No Presencial del Alumno/a	67,5		45	22,5					

M: Lecture-based

S: Seminar

TI: Industrial workshop

GA: Applied classroom-based groups GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-based groups GCA: Applied fieldwork groups

Evaluation methods

Legend:

- End-of-course evaluation

Evaluation tools and percentages of final mark

- Written test, open questions 80%

TA: Workshop

- Individual assignments 20%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

The written tests to be developed are detailed below:

A midterm exam will be held. The final exams will be attended with pending material.

The final mark of the exams will be the average of the two parts.

The deliverables to be carried out will consist of different tasks that will be described throughout the course, including the laboratory practices. Some should be done individually, others in groups. Some of them will be face-to-face and will take place in class.

In the event that presential evaluation of the subject cannot be carried out, the pertinent changes will be made to carry out an on-line evaluation by using the IT tools available at the UPV / EHU. The characteristics of this online assessment will be published in the student guides and in eGela

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

idem

MANDATORY MATERIALS

Teachers notes. Material available in egela

BIBLIOGRAPHY

Basic bibliography

Joseba García Melero. Resistencia de Materiales. Editorial: UPV-EHU

Detailed bibliography

Manuel Vazquez. Resistencia de Materiales. Editorial: Universidad Politécnica de Madrid Luis Ortiz Berrocal. Resistencia de Materiales. Editorial Mc Graw Hill Timoshenko. Resistencia de Materiales (2 tomos). Editorial: Espasa-Calpe

Journals

Web sites of interest

http://egela.ehu.eus



NAZIOARTEKO BIKAINTASUN CAMPUSA CAMPUS DE EXCELENCIA

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Univ del Paí

	IDE	2023/24	
Faculty	163 - Facult	y of Engineering - Vitoria-Gasteiz	Cycle .
Degree	GQUIIN10 -	Bachelor's Degree in Industrial Chemical Engineering	Year Third year
COURSE			
26091 - Ind	dustrial Chemi	stry	Credits, ECTS: 6
COURSE DE	SCRIPTION	•	
ENGLISH	VERSION		
		CHEMISTRY 26091	
AQCADEN	1IC YEAR 202	22/2023	
ASSOCIA	ED PROFES	SOR Dr.JOSE MARIA LOMAS	
Production		nation processes of raw materials and resources into the ma	ain chemical products.
CONTEXT			
Thermody Mechanics	namics, Q. Ino and Process	al, insofar as it applies the knowledge of other fields of Physiorganic, Q. Organic and Q. Physics, in addition to the field of Control . t collects elements of various subjects studied in the career,	Engineering, such as Fluid
OBJECTI∖	ES.		
Apply the k Know the r Visit variou Relate ene Minimize th Incorporate Promote the Relate prop	nowledge acc nanufacturing rgy and produ the damaging e the criteria o the developme duction proces	t chemical industrial production processes. quired in other subjects of the career in industrial reality, with methods of the main chemical products. Introduction to the ing facilities, with process monitoring on site. action aspects with their environmental impact. effects of large-scale material production f the "Commitment to Progress" of many of the large chemic ent of sustainable chemical manufacturing processes. asses to pollution. on data from the bibliography and specialized magazines of	economy of the sector.
SKILLS.			
2. Acquire 3. Underst 4. Be able	knowledge to and the transfe to design proc e basics of the	manage processes with material balances. TEQI1 manage processes with energy balances. TEQI1 ormation techniques of the main raw materials. TEQI2 cedures for the recovery of energy resources TEQI3 e management of manufacturing processes for different proc pplied experimentation procedures, equipment, and systems	ducts. TEQI2

11. Know, understand and apply the legislation, specifications, regulations and mandatory standards TEQI11

12. Make measurements, calculations, studies and reports, during the completion of each of the practices carried out in

NAZIOARTEKO BIKAINTASUN CAMPUSA CAMPUS DE EXCELENCIA INTERNACIONA

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the subject TEQI12.

METHODOLOGY

The presentation of the units is done in master classes, with an audio-visual support of graphics, figures and additional documents. These plugins are available on the "eGela" website.

Theoretical teaching is complemented with assistance to companies in the Chemical Sector, included as Field Practices. From them, reports are prepared on various industrial sectors, including economic and social aspects.

The subject is divided into two balanced parts.

Each part has a theoretical exam, developing several questions and / or problems.

Type of exam: descriptive questions of processes, reactions and applications of the substances studied.

Optional: According to the development of the course, a team work on a sector of the chemical industry will be proposed, to be presented orally and as a team.

TIMING AND TYPE OF TEACHING

M- Master class S- Workshop GCA-On site visits

NOTES

For each one of the theoretical topics taught, a Power Point document has been prepared, as well as other electronic documents that are uploaded on the Internet, through the eGela website, so that students can download it on their cell phone and / or personal computer.

In this subject it is intended that the student put into practice the knowledge acquired in the career, both in Chemistry and Engineering, for which it is considered that he must have passed most of the subjects that precede him in the curricular design.

COMPETENCIES/LEARNING RESULTS FOR THE SUBJECT

COMPETENCES / LEARNING OUTCOMES OF THE SUBJECT

The student who takes this subject will acquire the specific competences (CE) and transversal competences (CT) that are listed below.

These competencies are in line with the competencies referred to in the Degree Study Plan (BOE No. 30 of 02/04/2011), whose codes are specified in each case to facilitate their identification.

CODES: [C]Competences typical of the "Degree in Industrial Chemical Engineering" degree; [CRI] competencies common to the "Industrial Branch Engineering"; [TEQI] competencies of the "Industrial Chemistry Module".

SPECIFIC COMPETENCES (CE)

- 1. Be able to design and manage processes with material balances. TEQI1
- 2. Acquire knowledge to manage processes with energy balances. TEQI1
- 3. Understand the transformation techniques of the main raw materials. TEQI2
- 4. Be able to design procedures for the recovery of energy resources TEQI3
- 5. Learn the basics of the management of manufacturing processes for different products. TEQI2

6. Design and manage applied experimentation procedures, equipment, and systems management, relating thermodynamic concepts in physical processes TEQI5

7. Develop capacities and acquire skills to design compound synthesis processes applying the TEQI7 safety standards.

8. Acquire the ability to apply the strategies of scientific methodology: propose hypotheses solutions to solve problems of I. Chemistry - TEQI8

9. Be able to adequately communicate knowledge, procedures and results in the field of chemical engineering, using the specific vocabulary and terminology TEQI9.

10. Work effectively in multidisciplinary environments integrating skills and knowledge to make decisions in the field of chemical engineering TEQI10

 Know, understand and apply the legislation, specifications, regulations and mandatory standards TEQI11
 Make measurements, calculations, studies and reports, during the completion of each of the practices carried out in the subject TEQI12.

TRANSVERSAL COMPETENCES (CT)

Each one of the transversal competences (CT) corresponds to the competences [C], [CRI] and [TEQI] of the Study Plan that are indicated in each case.

•(CT1) Be able to rigorously use the appropriate terminology to adequately communicate knowledge and express oneself correctly in oral debates and in technical reports in this field [C4 / C5 / CRI14 / TEQI9 / TEQI12]. •(CT2) Be able to understand (in Spanish and English), interpret and question new scientific-technical information from bibliographic resources - of different types and formats -, developing an interest in learning and the ability to do so autonomously [C10/C12].

•(CT3) Adopt a responsible and orderly attitude, both in individual and cooperative work [C12/CRI16]

Theoretical and Practical Contents

SUMMARY.

The Chemical Industry. Raw Materials. Water as raw materials. Industry of alkali halides and soda. Nitrogen industry. Industry derived from phosphorus. Fertilizers Industry derived from sulfur. Auxiliary industries of construction. Glass and ceramic industry. Oil and its technology. Petrochemical industry. Plastics industry. Rubber and derivatives. Paper industry.

EXPOUNDED SUMMARY.

Unit 1- The Chemical Industry. Raw Materials

Historical development of the Chemical Industry. Basic considerations of the Chemical Industry. Economic importance of the world and Spanish Chemical Industry. Raw Materials in the Chemical Industry. Classification according to its origin. Worldwide distribution of raw materials and their consumption. Types of chemical transformations. Most important commercial chemicals.

Unit 3- Alkaline Halide Industry and Sosa

Separation of dissolved salts. Sodium chloride and derived salts. Electrolysis. Leblanc process. Solvay process. Sodium carbonate and its applications. Obtaining chlorine. Applications of chlorine and soda. Production of sodium. Potassium salts. Sylvinite treatments. Potassium chloride applications.

Unit 4- Nitrogen Industry.

Manufacture of ammonia: Raw materials. Synthesis of ammonia. Obtaining nitric acid by oxidation of ammonia. Nitric acid applications. Other products derived from ammonia: Manufacturing processes and applications.T

Unit 5- Industry derived from Phosphorus. Fertilizers

Phosphate rock as a raw material. Thermal Process: Obtaining elemental phosphorus and calcined phosphates, oxidation of phosphorus and production of thermal acid. Wet decomposition: Manufacture of superphosphates and phosphoric acid. Applications of phosphoric acid and derivatives.

Fertilizers: Nutrient elements of plants. Composition of cultivation soils, progressive depletion of the land, supply of nutrients and amendments. Fertilizers: NPK chemical fertilizers and balanced according to Patart. Formulation and metabolism of fertilizers. Manufacture of ternary fertilizers: Dosing of raw materials.

Unit 7- Auxiliary industries of construction.

Plaster: Transformation by thermal means and putting into work. Derivative products. Limestone: Obtaining lime and use as a binder. Portland cement: Characteristics and applications. Obtaining and composition of clinker. Manufacture of cement and commissioning. Other types of cements.

Unit 6- Industry derived from Sulfur. Raw materials for obtaining sulfur and its derivatives. Pyrite roasting: Use of gases and ashes. Manufacture of sulfuric acid: Catalytic oxidation of SO2, absorption of SO3. Sulfuric acid applications and obtaining its derivatives.

Unit 8- Glass and ceramic industry.

Silica as a raw material. Characteristics and chemical composition of glass. Manufacture of glass. Clay as a raw material. Properties and chemical composition. Manufacture of ceramic materials. Ceramic Industry Products: Characteristics and applications. Other applications of clay.

Unit 10- Oil and its technology.

Origin, extraction and "in situ" treatment of oil. Manufacturing processes in refinery. Initial distillation. Transformation processes of molecules (reforming, cracking, dehydrogenation ...). Synthesis processes (alkylation, polymerization, hydrogenation). Manufacturing schemes. Product debugging. Product applications. Lubricants.

Unit 12- Petrochemical Industry

Starting materials, variety of products and fields of application. Basic petrochemical techniques (separation of species, structural transformation of hydrocarbons: decomposition of hydrocarbons ...). Obtaining synthesis gas, acetylene, olefins

and aromatic compounds. Applications and derivative products

Unit 13- Polymer Industry. Rubber and derivatives

General concepts. Compounds involved: Polymers, fillers, reinforcements and additives. The solid state of polymers: glass transition temperature, melting temperature and other conditioning factors. Synthesis, addition, and condensation polymers. Thermoplastic and thermosetting materials. Transformation of plastics: compression, injection, extrusion. Tree extractable matter. Natural rubber technology: Obtaining, vulcanization process and applications. Manufacture of artificial rubbers. Rubber transformation techniques. Current trends in the elastomer sector.

Unit 14- Paper Industry

Wood constitution. Use of wood: Chemical transformation of cellulose, hydrolysis, pyrolysis. Paper technology: Methods for obtaining chemical pulps. Pastes conditioning. Paper manufacturing

TEACHING METHODS

METHODOLOGY.

MASTER CLASSES.

CE competences will be worked fundamentally.

• Classroom activity: The lecturer will explain the theoretical content and discuss application issues. The students will cooperatively resolve the issues raised by the teacher.

• Non-attendance activity: The student will individually work on the theoretical contents and the questions that are given as work material in each topic (self-assessment).

SEMINARS.

They will be used to carry out the activities related to the part of the program that is developed through the "Problem-Based Learning Methodology" (Items 1, 6, 7 and 8).

Classroom activity: The teacher will present the problem and guide the students in its analysis and resolution. The CE and CT competences indicated in each case will be worked on. Students will analyze the problem, identify learning objectives, and plan assignments.

Non-contact activity: students will carry out the planned tasks to achieve the learning objectives.

CLASSROOM PRACTICES.

COOPERATIVE ACTIVITIES.

They will be carried out. The CE and CT competences indicated in each activity will be worked on. They will be oriented to: (a) the application of the theoretical-practical contents developed in the master classes and (b) to achieve the learning objectives necessary to solve the problem/sub-problems.

Classroom activity: The teacher will present the activity to be carried out. After the activity, each group will present their work.

Non-contact activity: Students will carry out the activity cooperatively. The product of each activity will be a deliverable (includes self-assessment report). There will be an oral presentation of the activity.

TUTORIAL ACTION.

Additional material will be provided to students who need to redirect self-study.

TYPE OF TEACHING.

M- Master class S- Workshop

NOTES

For each of the theoretical Units taught, a Power Point document has been prepared, as well as other electronic documents that are uploaded on the Internet, through the eGela website, so that students can download it on their cell phone and / or personal computer.

In this subject it is intended that the student put into practice the knowledge acquired in the career, both in Chemistry and

Engineering, for which it is considered that he must have passed most of the subjects that precede him in the curricular design.

GCA-On site visits

TYPES OF TEACHING

Types of teaching	М	S	GA	GL	GO	GCL	ТА	TI	GCA
Hours of face-to-face teaching	45	5							10
Horas de Actividad No Presencial del Alumno/a	67,5	7,5							15

 Legend:
 M: Lecture-based
 S: Seminar

 GL: Applied laboratory-based groups
 GO: Applied computer-based groups

 TA: Workshop
 TI: Industrial workshop

GA: Applied classroom-based groups GCL: Applied clinical-based groups GCA: Applied fieldwork groups

Evaluation methods

- End-of-course evaluation

Evaluation tools and percentages of final mark

- Written test, open questions 70%
- Individual assignments 20%
- Teamwork assignments (problem solving, Project design) 10%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

ORDINARY EXAMS CALL: GUIDELINES AND RESIGNATION (SPANISH). EVALUATION SYSTEM.

This course offers two assessment systems: (A) CONTINUOUS ASSESSMENT or (B) FINAL EXAM. Students are encouraged to use the continuous assessment system to optimize the learning process and the acquisition of skills.

The lecturer will be informed in writing, prior to week 24, if the chosen evaluation system is option (B).

(A) CONTINUOUS ASSESSMENT (a minimum attendance of 60% is required for theoretical-practical classes) Includes WRITTEN TESTS and COOPERATIVE ACTIVITIES.

On site practices: Individual works, related to Field Practices: 1 point / 10

To assess the work on the visit it is necessary to have attended. Each job is graded 0-10.

Team work: The work done and its presentation are valued: 1 point.

Taking two theoretical-practical exams, corresponding to two parts of the subject: 7 points / 10

It is necessary to obtain a minimum of 4 points / 10 in the mark of each exam to pass. In that case, the average of both is taken

Class participation: 0.5 point / 10.

To consider class participation, it is required:

Regular

The Final Note corresponds to

Average mark of the partial or final exams: 75%

Note of field practical work: 20%

Class Participation Note: 5%

The evaluation criteria of the detailed previous aspects are published in eGela-Subject Conditions.

Conditions to pass the subject in the final exam in June or July

4 ≤ EXAM grade

FINAL NOTE = (∑ n PARTIAL EXAMS / n) x 0.75 + (∑ n WORKS / n) x 0.2 + Class participation. In case of not passing the subject, the works are saved for the following course only.

(B) FINAL EXAMINATION

The student must do:

(a) One or two written tests (50% of the final grade). They will include questions and application exercises. Specific skills will be evaluated.

(b) A practical activity (50% of the final mark) - proposal of technological alternatives for the management of RTPs - An oral presentation will be made to evaluate the transversal competences. It will be necessary to obtain a minimum score of 4/10 in (a) and (b) to mediate and pass the subject (5/10).

It will be necessary to obtain a minimum score of 4/10 in (a) and (b) to mediate and pass the subject (5/10). RESIGNATION OF CALL:

By writing to the teacher up to 1 month before the end of the teaching period.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

EXTRAORDINARY CALL: GUIDELINES

The student must do:

(a) A written test (50% of the final mark). It will include questions and application exercises. Specific skills will be evaluated.

(b) A practical activity (50% of the final mark) - proposal of technological alternatives for the RTP management -. An oral presentation will be made to evaluate the transversal competences. It will be necessary to obtain a minimum score of 4/10 in (a) and (b) to mediate and pass the subject (5/10).

MANDATORY MATERIALS

BIBLIOGRAPHY

Basic bibliography

Vian Ortuño, Angel. Introducción a la Química Industrial. Ed. Reverté, Barcelona, 2007. Riegel, E.R. Handbook of Industrial Chemistry. Riegel´s Handbook of Industrial Chemistry. 2004. Shreve, R. N., Austin, G.T. Chemical Process Industries (5º de.). Mc.Graw Hill, Nueva York. 2000.

Detailed bibliography

Stocchi, E. Industrial Chemistry. Ellis Horwood, Nueva York. 2010

Vincent Vela, María y col. Química industrial orgánica. Universidad Politécnica de Valencia. Servicio de Publicaciones. 2006

Maria R. Gómez Antón y col. Química Inorgánica y orgánica de interés industrial. Madrid 200

Journals

Ingeniería Química http://www.rbi.es/publicaciones/ingenieria-quimica.htm

Tecnología del Agua http://www.rbi.es/publicaciones/tecnologia-agua.htm

Residuos http://www.rbi.es/publicaciones/residuos.htm

Web sites of interest

Chemical Engineering. http://www.rbi.es/publicaciones/ingenieria-quimica.htm Water Technology. http://www.rbi.es/publicaciones/tecnologia-agua.htm webs de interés Federation of Chemical Industries of Spain. http://www.feique.org/ Main companies in the sector. http://www.quimicainfo.com/

OBSERVATIONS

Los trabajos de las prácticas de campo, en caso de no superar la asignatura, se guardan para el siguiente curso sólo un año.

COURSE GUIDE 2023/24				
Faculty 163 - Faculty of Engineering - Vitoria-Gasteiz	C	Cycle].	
Degree GQUIIN10 - Bachelor's Degree in Industrial Chemical Engineering	Y	ear	Fourth yea	ar
COURSE				
26095 - Industrial Waste Management		Credi	ts, ECTS:	6
COURSE DESCRIPTION				
SHORT DESCRIPTION.				

Environmental problem of industrial waste. Regulatory regulations. Conventional and advanced physicochemical technologies for the treatment of toxic and dangerous waste. Thermal technologies. Stabilization and controlled deposit. Minimization and valorization as management strategies. Forest, Agricultural and Solid Urban Waste and its treatment.

CONTEXTUALIZATION OF THE SUBJECT IN THE DEGREE.

This is an optional subject in the 4th year (2nd semester) that tries to provide the scientific-technological foundations of the integral management of industrial residues.

This subject does not have prerequisites, although it is advisable to have previously taken the subjects of the Industrial Chemistry Module, especially "Industrial Chemistry" in the third year and "Environmental Technologies", in the first semester of the 4th year.

The knowledge and skills acquired by studying "Industrial Waste Management", together with those of the optional subjects "Biological Treatment of Effluents" and "Treatment of Atmospheric Pollution", provide the technological foundations required for the comprehensive management of industrial waste (solid) and effluents (liquid and gaseous) generated in industrial activity.

The expertise and abilities related to the management of industrial residues are very useful in the professional practice of all branches of Industrial Engineering, since the activities of the most important industrial sectors generate toxic and dangerous waste.

COMPETENCIES/LEARNING RESULTS FOR THE SUBJECT

GENERAL OBJECTIVE

Provide the scientific-technological foundations of the main strategies to carry out an integral and sustainable management of toxic and dangerous industrial waste and know how to use the regulations that regulate its categorization and management.

The graduate in Industrial Engineering must know how to deal with the problems associated with the generation and management of this waste and be trained to promote minimization and recovery in production lines. Studying this subject acquires the basic training necessary for it.

COMPETENCES / LEARNING OUTCOMES OF THE SUBJECT.

The student who takes this subject will acquire the specific competences (CE) and transversal competences (CT) that are listed below.

These competencies are in line with the competencies referred to in the Degree Study Plan (BOE No. 30 of 02/04/2011), whose codes are specified in each case to facilitate their identification.

CODES: [C]Competences typical of the "Degree in Industrial Chemical Engineering" degree; [CRI] competencies common to the "Industrial Branch Engineering"; [TEQI] competencies of the "Industrial Chemistry Module".

SPECIFIC COMPETENCES (CE).

Each of the specific competencies (CE) corresponds to the competencies [C], [CRI] and [TEQI] of the Study Plan indicated in each case.

•(CE1) Be able to categorize the toxic and/or dangerous nature of the waste generated in industrial activities and assess its environmental management in accordance with current legislation [C6 / C11 / CRI17 / CRI18 / TEQI11]. •(CE2) Be able to decide the most suitable equipment and its characteristics for the different stages that, in general, make up the treatment of industrial waste – physical-chemical, thermal and stabilization-controlled deposit – using sustainability criteria [C3 / C5 / CRI10 / CRI18 / TEQI1 / TEQI12].

•(CE3) Be able to propose different treatment alternatives for the main types of industrial waste, sequentially describing all its stages, and the technological and sustainability advantages and disadvantages of each alternative [C4 / C5 / C7 / CRI10 / CRI13 / CRI18 /TEQI1/TEQI8/TEQI12].

•(CE4) Be able to question, decide and reasonably justify which is the Best Available Technology(s) (BAT) among

the different technological alternatives for the management of the main types of RTPs, using strategies of the scientific methodology and sustainability criteria [C3 / C4 / C5 / C7 / CRI10 / CRI13 / TEQI1 / TEQI8].

TRANSVERSAL COMPETENCES (CT),

Each one of the transversal competences (CT) corresponds to the competences [C], [CRI] and [TEQI] of the Study Plan that are indicated in each case.

•(CT1) Be able to rigorously use the appropriate terminology to adequately communicate knowledge and express oneself correctly in oral debates and in technical reports in this field [C4 / C5 / CRI14 / TEQI9 / TEQI12].

•(CT2) Be able to understand (in Spanish and English), interpret and question new scientific-technical information from bibliographic resources - of different types and formats -, developing an interest in learning and the ability to do so autonomously [C10/C12].

•(CT3) Adopt a responsible and orderly attitude, both in individual and cooperative work [C12/CRI16]

Theoretical and Practical Contents

THEORETICAL-PRACTICAL CONTENT.

CT: Theoretical Content CP: Practical Content.

UNIT 1- GENERALITIES, CONCEPTS AND ORIGIN OF WASTE.

Generalities, concepts and origin of waste. Environment. Contaminating vectors. Introduction to air pollution. Introduction to water pollution. Waste definitions. Zero residue notion. The limits of recycling. Recycling and materials science. The sustainability. Common environmental problems in the EU. Notion of Industrial Ecology

SUBJECT 2- CLASSIFICATION AND MANAGEMENT OF WASTE. THE BAG OF BY-PRODUCTS.

Waste treatment policy. Eco-production: prevention at source and minimization of waste. Minimization of containers and packaging. The Waste Bags. Recycling. The Waste Catalog. Waste classification. Introduction to the Life Cycle. Priorities in waste management. Spill and Life Cycle. Agenda 21.

UNIT 3- TECHNOLOGIES APPLICABLE TO WASTE TREATMENT. RECOVERY AND MANUFACTURING OF MATERIALS FROM WASTE.

Introduction and definitions related to waste treatment. Selection of the treatment process. Technologies applicable to multidisciplinary recycling. physical techniques. chemical techniques. Physicochemical techniques. Biological techniques. mixed systems. Regeneration of used solvents. Volatile compound treatments. Metal recovery treatments. The composting. Aerobic treatment process. The methanation. Aerobic treatment process. Summary of technologies suitable for recycling waste. Types of waste and technologies applicable to recycling.

UNIT 4- TYPOLOGY OF WASTE IN ORDER OF RECYCLING..

Notion of contaminant load. Risk factors associated with waste Leaching and contaminant load. Leaching test. resistance to leaching. Waste coding. Waste catalog. The typologies of waste, its coding and the systematics of the catalog for the treatment and recovery of waste. Potentially Hazardous Waste Constituents. Activities that can generate potentially hazardous waste. Waste and applicable technologies for its recycling. Contaminant concentration and recovery. Direct and indirect recycling. Multidisciplinary recycling. Recycling of materials in Europe. Contaminated soils as recoverable waste. Market for environmentally correct products and services. Installations, mechanisms and tools to reduce the impact of waste

UNIT 5- WASTE DESTINED FOR THE MANUFACTURE OF INSULATING MATERIALS.

Generalities, concepts and properties derived from light waste. Waste applications for the manufacture of insulators. Substrates and supports. Insulators made from waste. Acoustic screens

UNIT 6- WASTE DESTINED FOR THE MANUFACTURE OF DENSE MATERIALS.

Introduction. Contributions of ceramics to the recovery of waste. Recovery of waste with high asbestos content. The cement industry as a means of recovery of waste. The cement industry as a means of recovering waste



Vitrifiable waste. Inertization of waste through vitrification. Vitrification as an industrial technology. Vitrification of radioactive materials.

UNIT 8- ASSESSMENT OF WASTE FROM LARGE INDUSTRIES.

Recovery of waste from the energy mining industry. The electricity generation sector. Recovery of waste from the steel and metallurgical industry. Chemical Industry

UNIT 9 - MINING WASTE.

NOT THIS YEAR

UNIT 10- USE OF AGRICULTURAL AND FORESTRY RESIDUES.

Definitions and introduction. Origin and sources of biomass. Strategies for Energy recovery from biomass. Biomass utilization potential and costs.

UNIT 11 VALORIZATION AND TREATMENT OF MUNICIPAL WASTE.

Introduction. Composition and evolution of MSW. Toxic waste in MSW. Controlled landfill

UNIT 12. SLUDGE FROM WWTP TREATMENT PLANT.

NOT THIS YEAR

UNIT 13. RESIDUAL PLASTICS AND THEIR REUSE.

NOT THIS YEAR

TEACHING METHODS

METHODOLOGY.

MASTER CLASSES.

CE competences will be worked fundamentally.

• Classroom activity: The lecturer will explain the theoretical content and discuss application issues. The students will cooperatively resolve the issues raised by the teacher.

• Non-attendance activity: The student will individually work on the theoretical contents and the questions that are given as work material in each topic (self-assessment).

SEMINARS.

They will be used to carry out the activities related to the part of the program that is developed through the "Problem-Based Learning Methodology" (Items 1, 6, 7 and 8).

Classroom activity: The teacher will present the problem and guide the students in its analysis and resolution. The CE and CT competences indicated in each case will be worked on. Students will analyze the problem, identify learning objectives, and plan assignments.

Non-contact activity: students will carry out the planned tasks to achieve the learning objectives.

CLASSROOM PRACTICES.

COOPERATIVE ACTIVITIES.

They will be carried out. The CE and CT competences indicated in each activity will be worked on. They will be oriented to: (a) the application of the theoretical-practical contents developed in the master classes and (b) to achieve the learning objectives necessary to solve the problem/sub-problems.

Classroom activity: The teacher will present the activity to be carried out. After the activity, each group will present their work.

Non-contact activity: Students will carry out the activity cooperatively. The product of each activity will be a deliverable (includes self-assessment report). There will be an oral presentation of the activity.

TUTORIAL ACTION.

Additional material will be provided to students who need to redirect self-study.

TYPES OF TEACHING

Types of teaching	М	S	GA	GL	GO	GCL	TA	TI	GCA
Hours of face-to-face teaching	30	15	15						
Horas de Actividad No Presencial del Alumno/a	45	22,5	22,5						

M: Lecture-based S: Seminar

GL: Applied laboratory-based groups GO: Applied computer-based groups TI: Industrial workshop

GA: Applied classroom-based groups GCL: Applied clinical-based groups GCA: Applied fieldwork groups

Evaluation methods

Legend:

- End-of-course evaluation

Evaluation tools and percentages of final mark

- Written test, open questions 55%

TA: Workshop

- Teamwork assignments (problem solving, Project design) 45%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

ORDINARY EXAMS CALL: GUIDELINES AND RESIGNATION (SPANISH).

EVALUATION SYSTEM..

This course offers two assessment systems: (A) CONTINUOUS ASSESSMENT or (B) FINAL EXAM. Students are encouraged to use the continuous assessment system to optimize the learning process and the acquisition of skills.

The lecturer will be informed in writing, prior to week 24, if the chosen evaluation system is option (B).

(A) CONTINUOUS ASSESSMENT (a minimum attendance of 60% is required for theoretical-practical classes) Includes WRITTEN TESTS and COOPERATIVE ACTIVITIES.

The written tests will account for 50% and the cooperative activities for 50% of the final grade for the subject. The FINAL MARK will be obtained by applying these percentages to the average grade of the written tests and to the average grade of the cooperative activities, with the following REQUIREMENTS:

(a) To average the note of the written tests it will be necessary to obtain a minimum grade of 4/10 in each one of them. (b) To average the note of the cooperative activities it will be necessary to have carried out all the activities and obtain a minimum grade of 4/10 in each one of them.

To pass the subject it will be necessary to obtain a final grade of 5/10.

The results of the evaluation will be reported within a week after taking the test or handing in and presenting the activity in class, so that the student can redirect learning by following the guidelines provided by the teacher.

WRITTEN TESTS (50%).

One or two written tests will be carried out and, if the result of the evaluation is negative, a third test (in June). The tests will include questions and application exercises related to the theoretical-practical contents that are the object of evaluation in each test.

The set of tests will evaluate the specific competences [CE1, CE2, CE3, CE4].

If the evaluation of the written tests is not favourable, the teacher will provide the student with complementary material to pass the learning objectives. Tutorials will be used to resolve doubts and/or difficulties that arise.

COOPERATIVE ACTIVITIES (50%).

Various cooperative activities will be carried out. Each of them will be assigned a score based on its degree of difficulty and the scope of the work. In each cooperative activity, the achievement of all TRANSVERSAL COMPETENCES will be evaluated, as well as the specific competence(s) that are indicated in each of them – depending on their theme and the type of activity-.

The evaluation instrument in each activity will be an EVALUATION RUBRIC OR MATRIX, which WILL DETAIL THE CRITERIA AND INDICATORS that are used to evaluate the achievement of the competencies and the SCORE of each of them, which will be used to analyze and assess the PRODUCTS OF ACTIVITY: (a) The DELIVERABLE generated in the activity and its ORAL PRESENTATION. The evaluation criteria will use as

indicators the structure of the work, the success and rigor in the treatment of the contents, the absence of misconceptions, the validity/feasibility of the proposed technological solution, the quality of the bibliographic sources used, the presentation and the quality of the writing, as well as the clarity of the oral presentation, the ability to communicate and the ability to respond in the debate, among others.

(b) The SELF-ASSESSMENT REPORT OF THE COOPERATIVE WORK CARRIED OUT. The evaluation criteria will use as indicators the degree of participation of the members of the group in the cooperative forum and the relevance of their contributions, their attitude in the group, the coordination of the members, the adjustment to the planned schedule and work plan, the participation in the oral presentation and/or in the debate and the degree of knowledge of the work carried out that is demonstrated in both cases, among others.

If the evaluation of the tasks is not satisfactory, the teacher will guide the group to modify the aspects of the activity that need to be improved, in order to redirect the learning and pass - if necessary - a new evaluation of the activity. Tutorials will be used to resolve doubts and/or difficulties and monitor the work.aluación de la actividad.

Tutorials will be used to resolve doubts and/or difficulties and monitor the work.

(B) FINAL EXAMINATION.

The student must do:

(a) One or two written tests (50% of the final grade). They will include questions and application exercises. Specific skills will be evaluated.

(b) A practical activity (50% of the final mark) - proposal of technological alternatives for the management of RTPs - An oral presentation will be made to evaluate the transversal competences.

It will be necessary to obtain a minimum score of 4/10 in (a) and (b) to mediate and pass the subject (5/10).

RESIGNATION OF CALL:

By writing to the teacher up to 1 month before the end of the teaching period.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

EXTRAORDINARY CALL: GUIDELINES.

The student must do:

(a) A written test (50% of the final mark). It will include questions and application exercises. Specific skills will be evaluated.

(b) A practical activity (50% of the final mark) - proposal of technological alternatives for the RTP management -. An oral presentation will be made to evaluate the transversal competences. It will be necessary to obtain a minimum score of 4/10 in (a) and (b) to mediate and pass the subject (5/10).

MANDATORY MATERIALS

RODRÍGUEZ JIMÉNEZ J.J. E IRABIEN GULIAS A. Gestión sostenible de residuos peligrosos. Madrid: Editorial Síntesis S.A., 2013.

CARMEN OROZCO BARRENETXEA ET AL. Contaminación Ambiental. Una Visión Desde La Química. Ediciones Paraninfo. 2011.

LAGREGA M.D. Gestión de residuos tóxicos. Tratamiento, eliminación y recuperación de suelos. New York: Mc Graw-Hill, 2006.

NEMEROW N.L. Tratamiento de vertidos industriales y peligrosos. Madrid: Díaz de Santos, 2006.

BIBLIOGRAPHY

Basic bibliography

LEVIN, M A [1]; GEALT, M A. Biotreatment of industrial and hazardous waste. U.S. Department of Energy. Office of Scientific and Technical Information. 2003

ELÍAS CASTELL . Reciclaje de residuos industriales: aplicación a la fabricación de materiales de construcción. Madrid: Díaz de Santos, 2012.

Detailed bibliography

PARDEEP SINGH, PRAMIT VERMA, et al.. Waste Management and Resource Recycling. Elsevier, 2022.

GEORGE TCHOBANOGLOUS, FRANK KREITH. Handbook of Solid Waste Management. McGraw HILL PROFESSIONAL, 2002 WOODARD & CURRAN, I. Industrial Waste Treatment Handbook. Elsevier, 2005

Journals

ASCE Practice Periodical of Hazardous Toxic and Radioactive Waste Management (USA). ASCE. http://www.pubs.asce.org/journals/hz.html

Environmental Law & Management. Wiley Interscience. http://www3.interscience.wiley.com/maintenance.html?DESCRIPTOR=PRINTISSN&VALUE=1067-6058

Environmental Science & Technology. American Chemical Society (ACS). http://pubs.acs.org/journals/esthag/index.htm

Residuos. La revista técnica de Medio Ambiente. http://www.revistaresiduos.com/RevistaResiduos/

Warmer Bulletin España. Instituto para la sostenibilidad de los Recursos (ISR). http://www.isrcer.org/warmer_new.asp

Waste Management. International Journal of Integrated Waste Management, Science and Technology. Elsevier. http://www.elsevier.com/wps/find/journaldescription.cws_home/404/description#description

Web sites of interest

Activities of the European Union in the field of waste management. Legislation. http://europa.eu.int/scadplus/leg/es/s15002.htm

Community of Waste Management Professionals. http://www.ictnet.es/ICTnet/home/cv/?area=mAmb&cv=residuos

ECOVIDRIO. Non-profit organization for glass recycling in Spain https://www.ecovidrio.es/

National Institute of Safety and Hygiene. Ministry of Labour and Social Affairs. Legislation. http://www.mtas.es/insht/legislation/tl_res.htm

Tecnociencia. Technical information on waste. Legislation. http://www.tecnociencia.es/especiales/residuos/

OBSERVATIONS

COURSE GUI	DE	2023/24											
Faculty	163 - Faculty of	Engineering -	Vitori	a-Gaste	eiz					Cycl	е	.	
Degree	GIAUTO10 - Ba	5 5				gineerir	ng - DU	AL		Year		Second ye	ear
OURSE						-	_			L		, ,	
28126 - Str	ructure design in	Automotive Er	nginee	ring							Credi	ts, ECTS:	6
	SCRIPTION												
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	Type Hours of face-to-f	es of teaching ace teaching	M 15	S	GA 30	GL 15	GO	GCL	ТА	TI	GCA	-	
Horas de Activ	vidad No Presencia	-	22,5		45	22,5							
Legend:	M: Lecture-based GL: Applied labora TA: Workshop	atory-based grou	ps GC	Seminar D: Applie Industria	d compu	iter-based	d groups	GCL:	Applied (assroom clinical-b fieldwork	ased gr	oups	
Evaluation m													
	ous evaluation												
- Continuo													
- Continuo - End-of-co	ols and percent	ages of final	mark										

Universidad Euskal Herriko del Pais Vasco Unibertsitatea

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

The written tests to be developed are detailed below:

A midterm exam will be held in the middle of the semester. The approval of this exam releases contents. The final exams will be attended with pending contents.

The final mark of the exams will be the average of the two parts.

Who does not appear to the final exam, will obtain a grade of not presented.

The work has a value of 50% of the final grade. For the evaluation of the work, the following will be taken into account: - The document: Quality, skills developed at work (minutes, tutorials, face-to-face hours).

- The presentation: Peer evaluation, teacher evaluation, attendance and participation.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

The same as in ordinary call.

MANDATORY MATERIALS

Theory and problems explained during lectures.

BIBLIOGRAFÍA

Basic bibliography

Joseba García Melero. Resistencia de Materiales. Editorial: UPV-EHU Joseba García Melero. De Leonardo da Vinci y Galileo a Mariotte. Editorial: Raima Joseba García Melero. De Parent y Coulomb a Navier y Saint-Venant. Editorial: Raima. Luis Ortiz Berrocal. Resistencia de Materiales. Editorial Mc Graw Hill

Detailed bibliography

Material curso OCW diseño de máquinas

SHIGLEY, S.E.: "Diseño en Ingeniería Mecánica".

Robert. L. Norton. "Diseño de Maquinaria".

Robert L. Mott "Diseño de Elementos de Máquinas".

Carlos Angulo, Luis Norberto López de Lacalle, Josu Agirrebeitia, Charles Pinto. "Elementos de Máquinas".

M.F. Spotts. "Proyecto de elementos de máquinas".

Journals

Web sites of interest

OBSERVATIONS

In the event that a face-to-face assessment of the subject cannot be carried out, the pertinent changes will be made to carry out an on-line assessment by using the existing IT tools at the UPV/EHU. The characteristics of this on-line assessment will be published in the student guides (annex or action plan) and in eGela.

COURSE GU	IIDE	2023/24											
Faculty	163 - Faculty of	of Engineering	- Vitori	a-Gaste	eiz					Сус	le].	
Degree	GIAUTO10 - E	Bachelor's Degi	ree in A	Automo	tive En	gineerir	ng - DL	JAL		Year	,	Second ye	ear
COURSE	1						<u> </u>						
28132 - Fi	nite Element Sirr	nulation and Ar	alisys	in Auto	motive	Engine	ering				Credi	ts, ECTS:	6
COURSE DE	SCRIPTION												
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		pes of teaching	М	S	GA	GL	GO	GCL	ТА	TI	GCA		
	Hours of face-to	-	15		15		30					_	
Horas de Acti	ividad No Presenci	al del Alumno/a	22,5		22,5		45						
Legend:		d ratory-based grou	ips GC	Seminar): Applie Industria	d compu	ter-base	d groups	GCL:	Applied	clinical-t	n-based based gr k group	oups	
Evaluation m	nethods												

- Exercises, cases or problem sets 40%

NAZIOARTEKO BIKAINTASUN CAMPUSA CAMPUS DE EXCELENCIA

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- Teamwork assignments (problem solving, Project design) 20%
- PRUEBA REALIZADA EN ORDENADOR 30%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

Se realizará evaluación continua en base al sistema de evaluación presentado.

En el caso de que no se pueda realizar una evaluación presencial de la asignatura, se realizarán los cambios pertinentes para la realización de una evaluación on line mediante la utilización de las herramientas informáticas existentes en la UPV/EHU. Las características de esta evaluación on line será publicadas en las guías de estudiante y en eGela.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

El alumnado tendrá que evaluarse de la parte que tiene suspendida.

En el caso de que no se pueda realizar una evaluación presencial de la asignatura, se realizarán los cambios pertinentes para la realización de una evaluación on line mediante la utilización de herramientas informáticas existentes en la UPV/EHU. Las características de esta evaluación on line será publicadas en las guías de estudiante y en eGela

MANDATORY MATERIALS

- Teachers notes

BIBLIOGRAFÍA

Basic bibliography

- Teachers notes

Detailed bibliography

-The Finite Element Method, O.C. Zienkiewicz. Ed Reverté.

Journals

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Web sites of interest

OBSERVATIONS

OURSE GUIDE	2023/24								
Faculty 163 - Facult	y of Engineering -	Vitoria-Gast	eiz			Cycl	е		
Degree GIAUTO10	- Bachelor's Degre	ee in Automo	otive Engineering	g - DUAL		Year		Fourth ye	ar
OURSE									
28141 - Advanced Autor	nation in Vehicle F	Production					Credit	s, ECTS:	4,5
COURSE DESCRIPTION									
Automation is a basic pil continuous innovation. In undertaken, through the	n this subject, the	integration o	f the different te	chnologies	structure	d in the	autom		
COMPETENCIES/LEARNI	NG RESULTS FO	R THE SUB	JECT						
The optional subject con delves into the acquisitic the automation of indust	n of knowledge ba			•			-	•	•
As a result of learning:									
 The student knows the 	Automotive Pyran	nid and its le	vels. He differer	ntiates the	function of	f each	level.		
 The student knows the 	concept of Indust	ry 4.0 and its	s characteristic e	elements.					
The student knows how	v to use the differe	ent software	tools that are int	egrated in	to an indu	strial au	utomati	on project	
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Hours of face-to-face teaching	15			30					
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M: Lecture-based	S:	Seminar				GA: A	pplied cl	assroom	i-based g
GL: Applied laboratory-based group	os GC): Applie	d comput	ter-based	d groups	GCL:	Applied	clinical-b	ased gro
TA: Workshop	TI:	Industria	al worksh	ор		GCA:	Applied	fieldwork	c groups
	ridad No Presencial del Alumno/a M: Lecture-based GL: Applied laboratory-based group	vidad No Presencial del Alumno/a22,5M: Lecture-basedS:GL: Applied laboratory-based groupsGC	idad No Presencial del Alumno/a22,5M: Lecture-basedS: SeminarGL: Applied laboratory-based groupsGO: Applied	idad No Presencial del Alumno/a22,5M: Lecture-basedS: SeminarGL: Applied laboratory-based groupsGO: Applied comput	vidad No Presencial del Alumno/a22,545M: Lecture-basedS: SeminarGL: Applied laboratory-based groupsGO: Applied computer-based	vidad No Presencial del Alumno/a 22,5 45 M: Lecture-based S: Seminar GL: Applied laboratory-based groups GO: Applied computer-based groups	vidad No Presencial del Alumno/a 22,5 45 M: Lecture-based S: Seminar GA: A GL: Applied laboratory-based groups GO: Applied computer-based groups GCL:	vidad No Presencial del Alumno/a 22,5 45 M: Lecture-based S: Seminar GA: Applied cl GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied cl	vidad No Presencial del Alumno/a 22,5 45 M: Lecture-based S: Seminar GA: Applied classroom GL: Applied laboratory-based groups GO: Applied computer-based groups GCL: Applied clinical-b

Universidad Euskal Herriko del Pais Vasco Unibertsitatea

Evaluation methods

- Continuous evaluation
- End-of-course evaluation

Evaluation tools and percentages of final mark

- Exercises, cases or problem sets 65%
- Teamwork assignments (problem solving, Project design) 35%

ORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

The default evaluation method is continuous evaluation, which will be graded as follows

Completion of practices (exercises, problems): 65%

Team work (projects): 35%

According to the Regulations governing the Evaluation of Students in official Degrees, chapter II, article 8, section 3, all students will have the right to be evaluated through the final evaluation system, regardless of whether or not they have participated in the continuous evaluation system. To do this, students must submit in writing to the teaching staff responsible for the subject the resignation to continuous assessment. Students will have a period of 9 weeks from the beginning of the semester, in accordance with the center's academic calendar to submit their resignation.

The Final Evaluation will consist of a written test and a practical exercise that will comprise 70% and 30% of the subject, respectively. It will be necessary to pass each of the parts.

EXTRAORDINARY EXAMINATION PERIOD: GUIDELINES AND OPTING OUT

According to the Regulations governing the Evaluation of Students in official Degree degrees, chapter II, article 9, section 2, this type of evaluation will be carried out exclusively through the final evaluation system. Likewise, according to section 3 of the aforementioned article, said test could consist of as many exams and evaluation activities as necessary.

MANDATORY MATERIALS

Given the high technological component of the subject and the great dynamism of the technologies it studies, the teaching team of the subject will indicate at the beginning of the same the material of obligatory use in it (if any).

BIBLIOGRAPHY

Basic bibliography

There is no invariably up-to-date reference bibliography. Therefore, following the methodology used in academic research, the student will be shown how to obtain a personalized bibliography, which is adapted to their specific needs and oriented towards their future professional career.

Detailed bibliography

In the same way, It is not possible to provide an invariable advanced bibliography. The student's training in the methodology of scientific research will allow him to obtain this bibliography in a personalized way and oriented to his professional interests.

Journals

The student's training in the methodology of scientific research will allow him to access scientific journals oriented to his professional interests.

Web sites of interest

Students will be trained to search for web links that are useful for their learning.

OBSERVATIONS